

# Ammonia Volatilization from Urea, Mechanisms, Magnitude, and Management

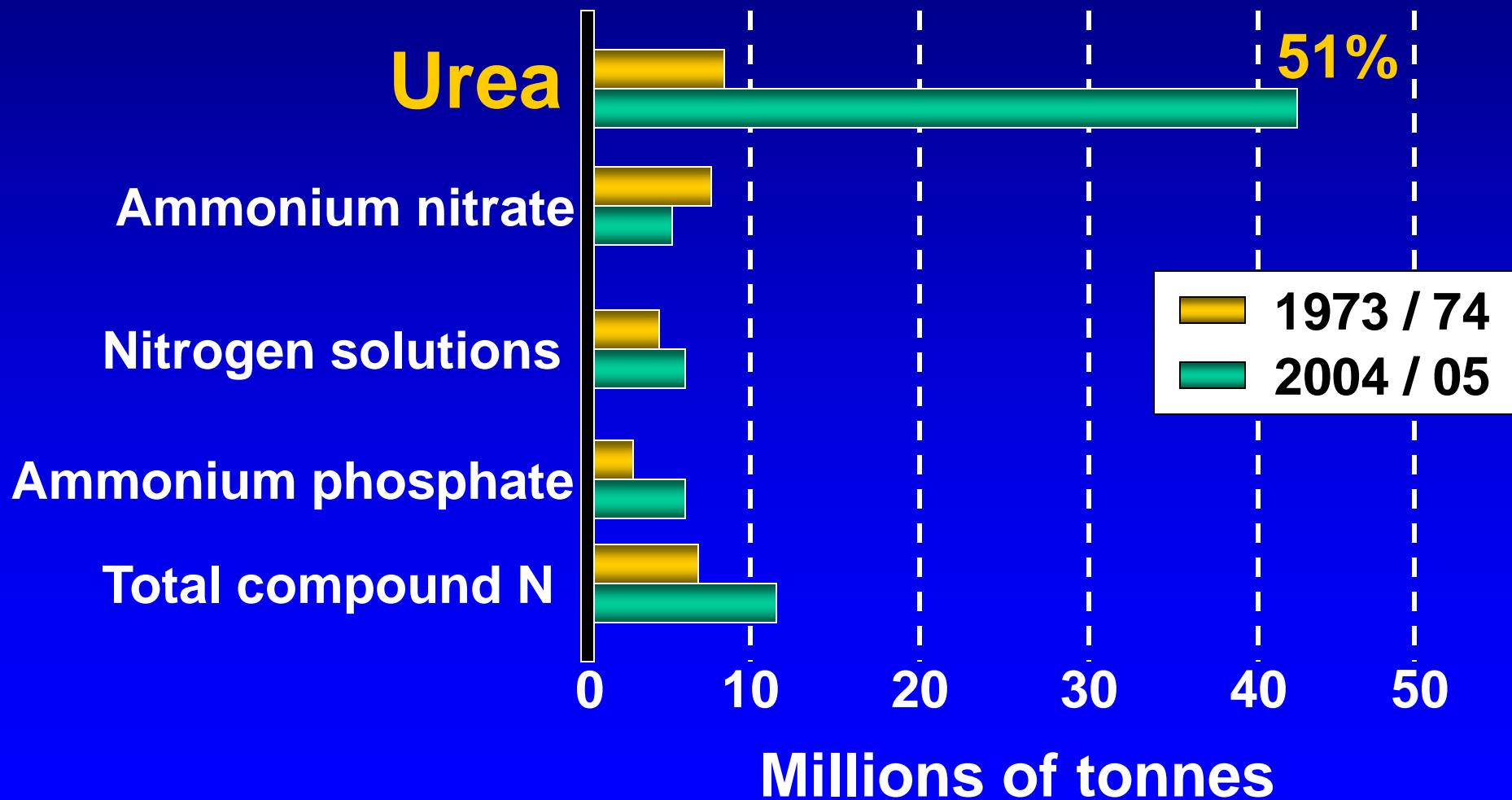
David E. Kissel

University of Georgia

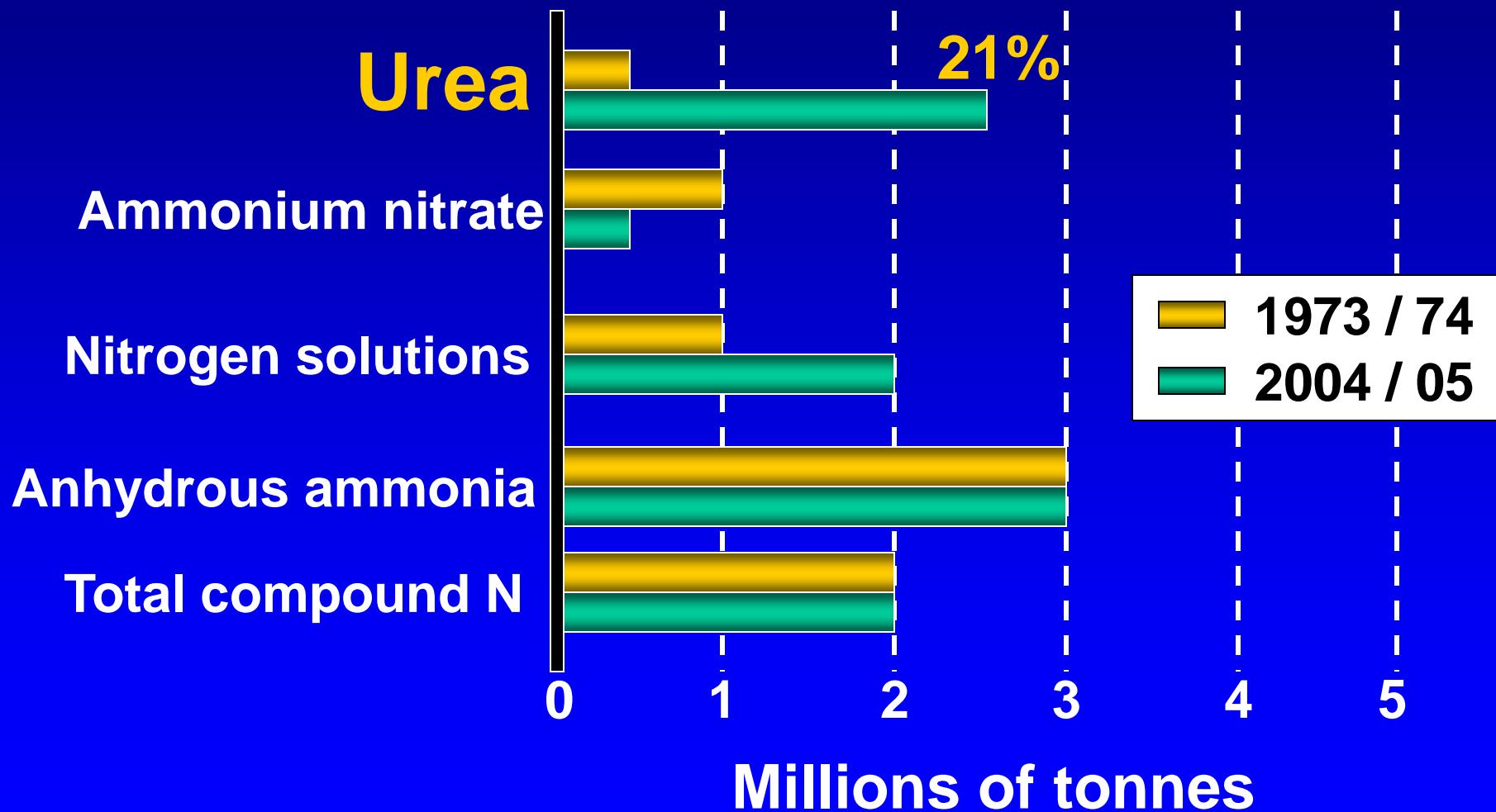
# Overview

- Urea Consumption
- Definitions
- UAN Solution Reactions
- Urea Dissolution and Diffusion
- Urea Hydrolysis
- Ammonia Volatilization
- Research Results
- Summary

# World Evolution of N Fertilizer Consumption



# USA Evolution of N Fertilizer Consumption



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# Definitions

- Dissolution: urea absorbs water and converts from solid to liquid
- Hydrolysis: urea converts to ammonium ( $\text{NH}_4^+$ )
- Diffusion: movement due to motion of molecules
  - Urea has no charge, diffuses easily
  - $\text{NH}_4^+$  has charge, diffuses slowly

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Q. Does any ammonia volatilize from UAN solutions before they are applied or during the application?

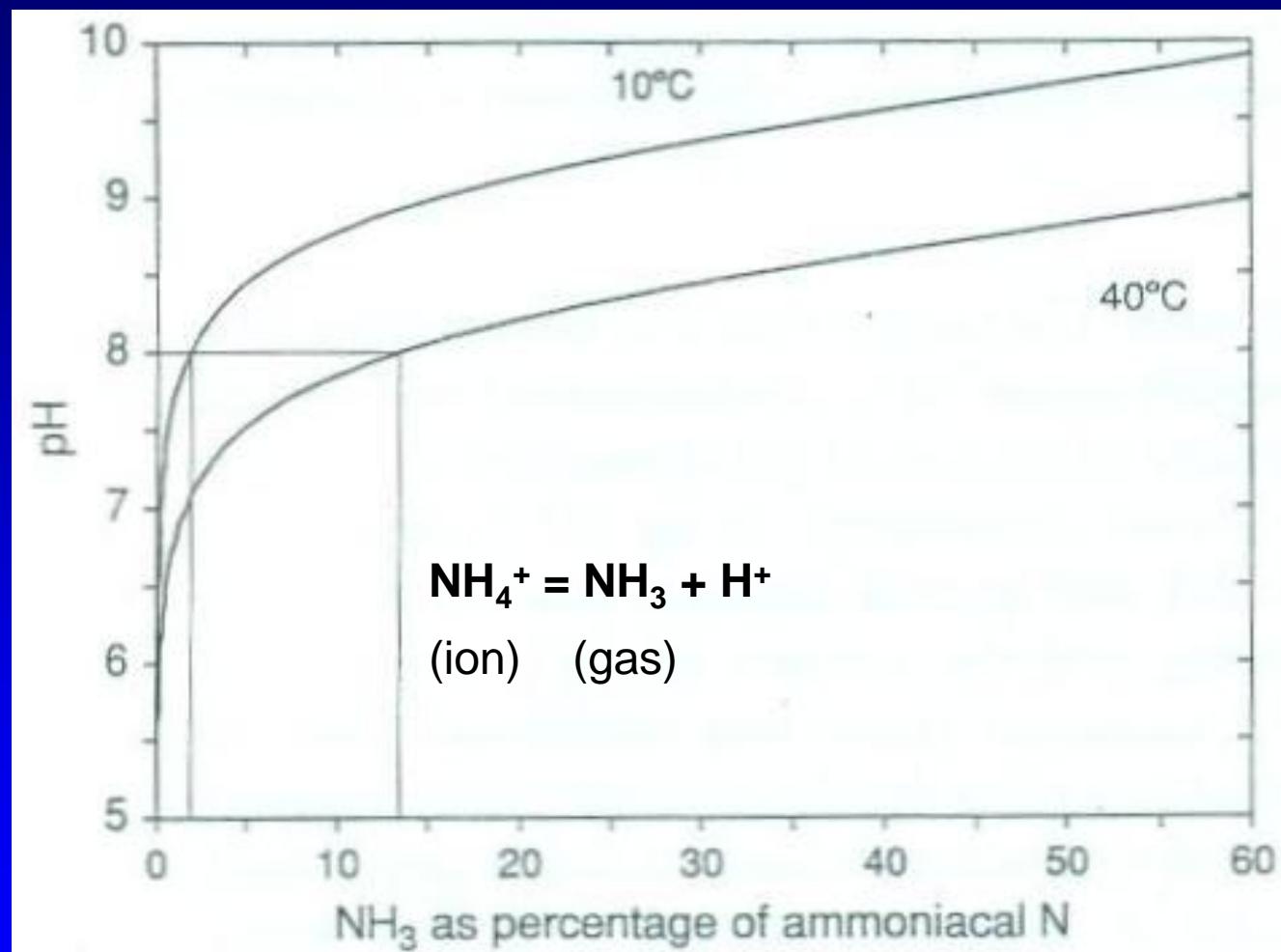
A. None or very little, depending on if a small amount of ammonia is added to protect against corrosion of mild steel.

# Composition of UAN 28% N

- 14 % N from Ammonium Nitrate
  - 7% ammonium N
  - 7% nitrate N
- 14% N from Urea
- 30% water
- + small amounts of inhibitors to inhibit corrosion of mild steel

# Inhibitors to protect against corrosion of mild steel

- 0.5 % ammonia (raises pH to about 7.5)
- Ammonium phosphates at 0.2 %  $P_2O_5$
- Others



# Urea Hydrolysis in UAN?

- DOES NOT OCCUR
  - Therefore no ammonia is formed from the urea portion of the UAN solution from this process.
  - Therefore no ammonia loss from the urea portion of UAN solution before soil application.

# Summary of ammonia loss from UAN

- Urea in UAN does not hydrolyze in the fertilizer tank.
- The  $\text{NH}_4^+$  from the ammonium nitrate portion of the UAN cannot be lost as  $\text{NH}_3$ .
- The amount of  $\text{NH}_3$  added to some UAN to inhibit corrosion is very small, around 10 lb per ton. A small portion of this  $\text{NH}_3$  may be lost during application.

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# Urea Dissolution

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- Urea can absorb water from the atmosphere and from the soil/crop residue

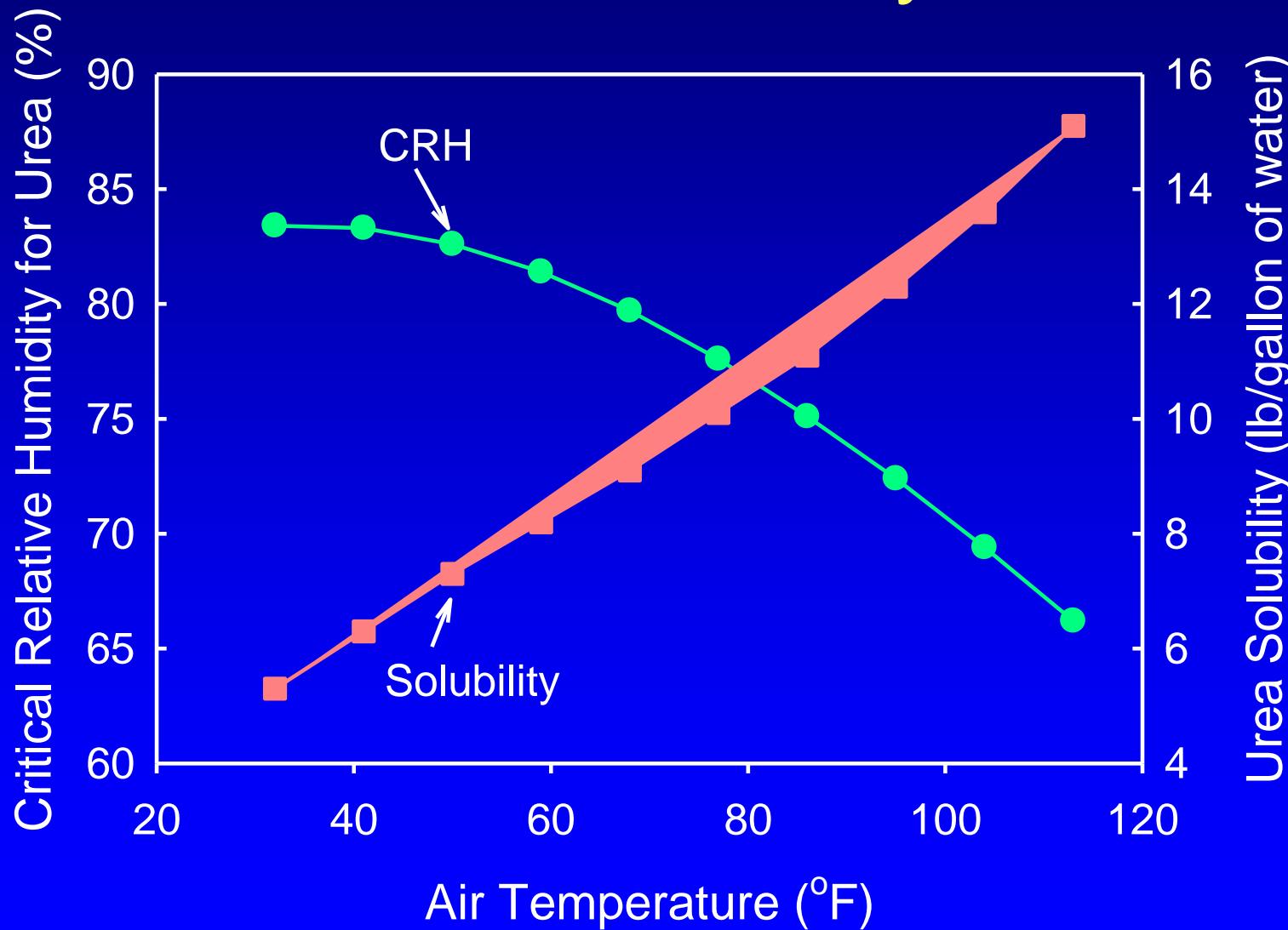
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- Urea has to be dissolved before it can be hydrolyzed
- Urea can absorb water from the atmosphere and from the soil/crop residue
- Absorption of water from the atmosphere depends on relative humidity (RH)
- Critical relative humidity (CRH) is the RH at which urea dissolves

# Effect of Temperature on CRH and Water Solubility of Urea

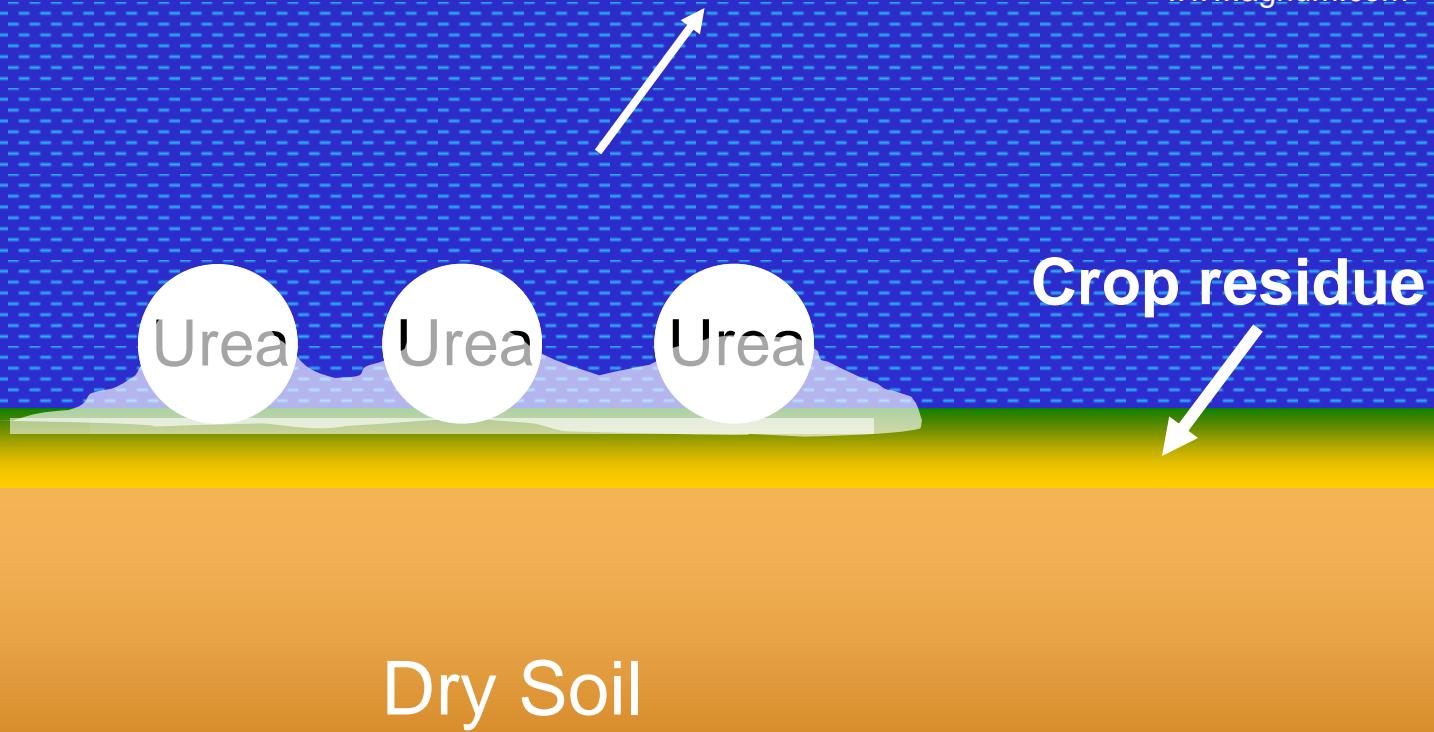


# Urea dissolution and diffusion

$\text{RH} > \text{CRH} (80\%)$



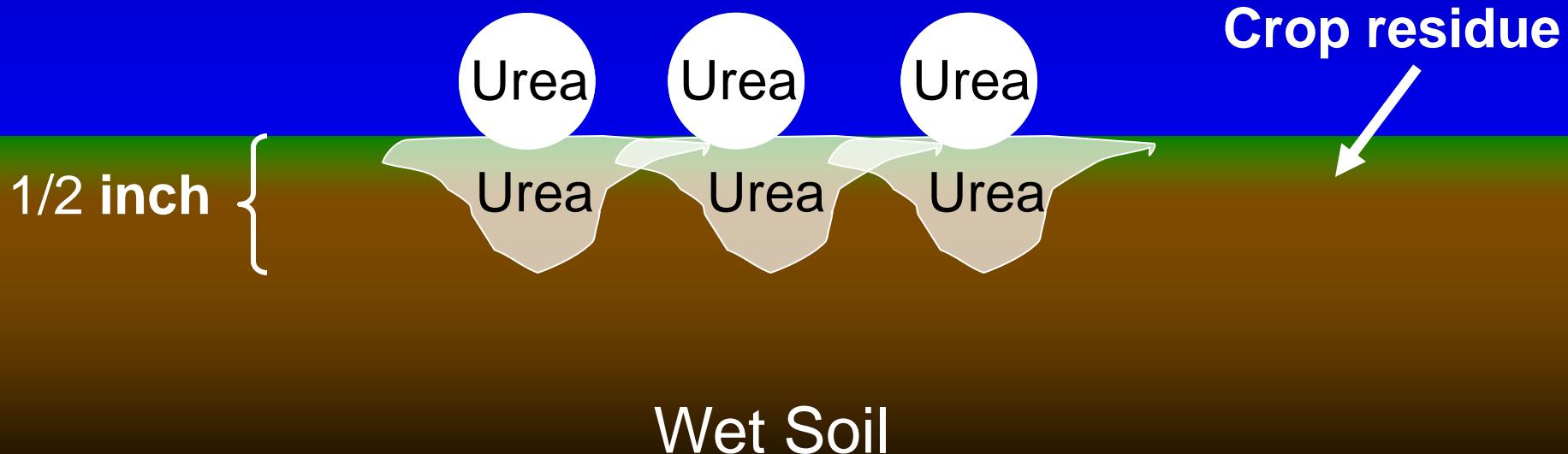
[www.agrium.com](http://www.agrium.com)



# Urea dissolution and diffusion

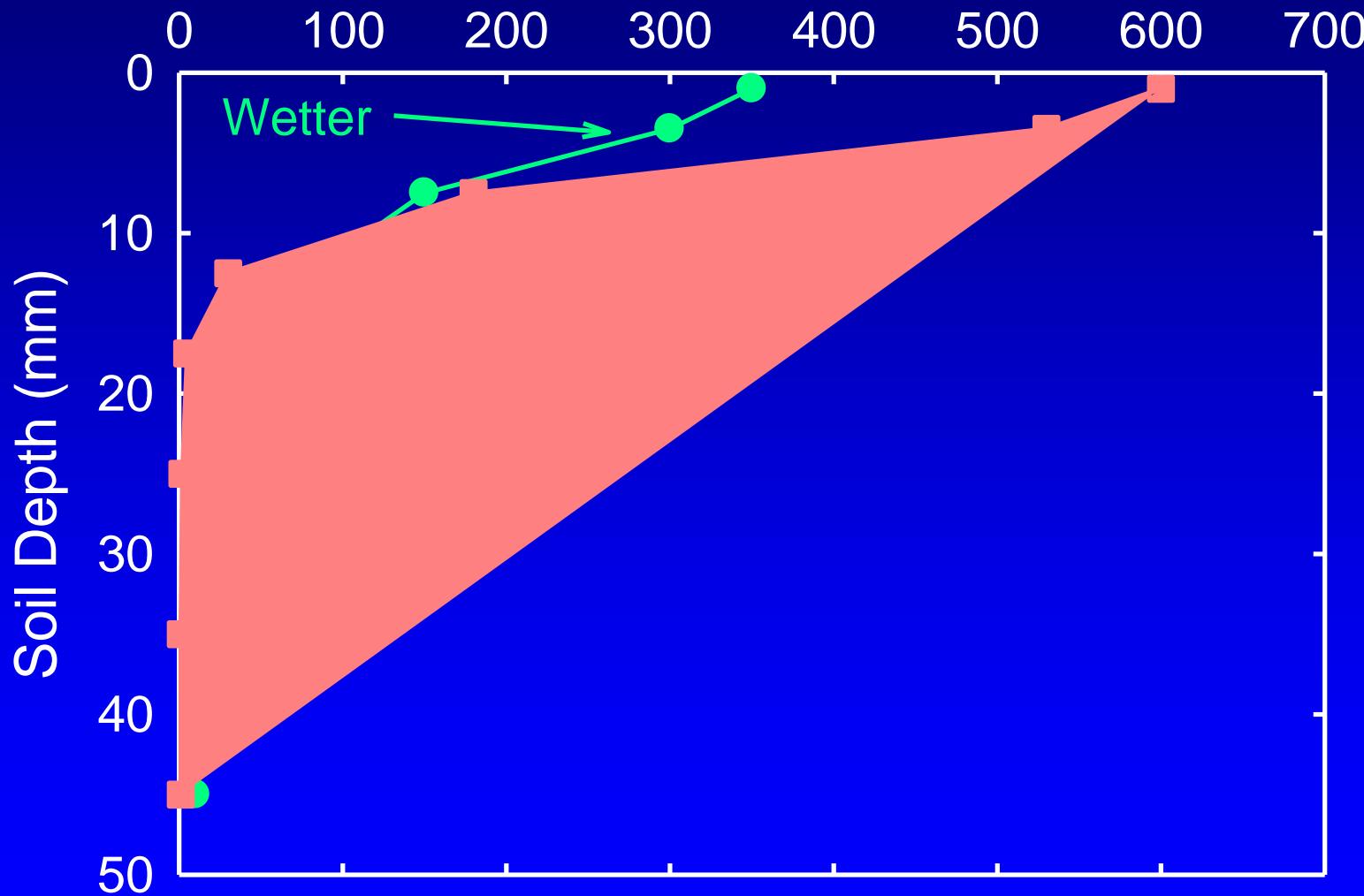


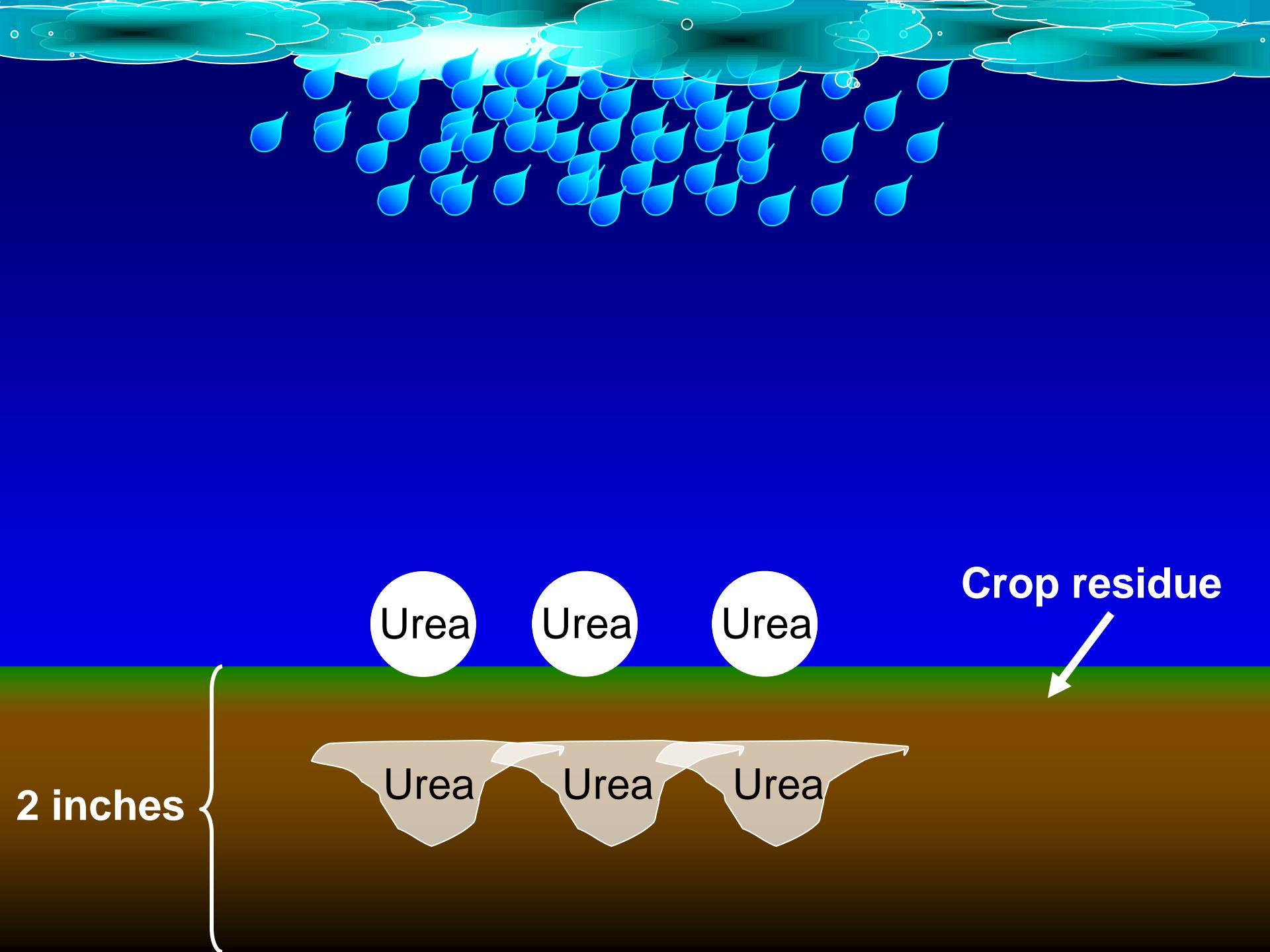
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# Urea Diffusion into Soil (after 10 days)

Urea Concentration in Soil (ppm)





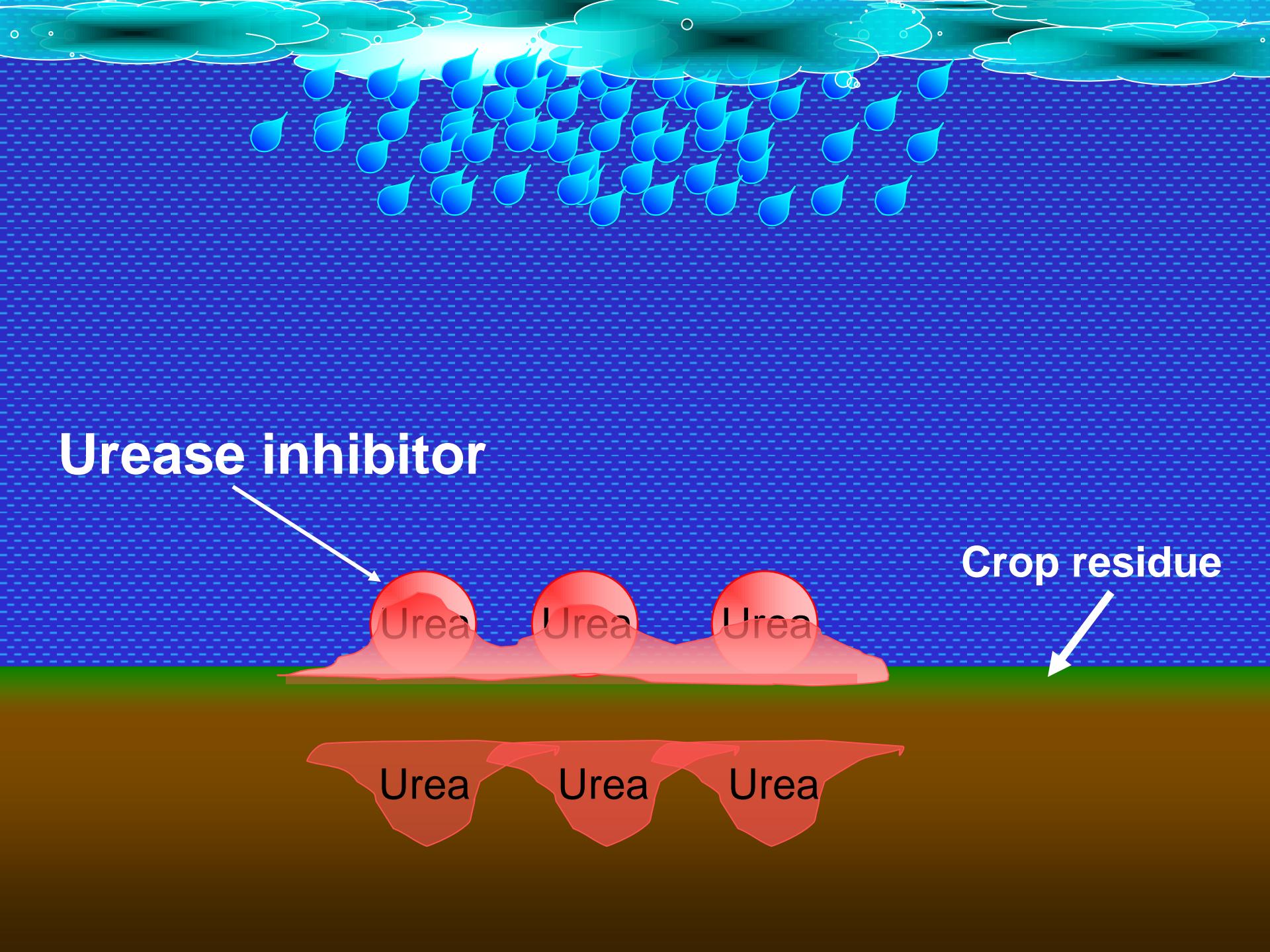
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# Urea Hydrolysis



- Urease is derived from crop residues and soil microorganisms



Urease inhibitor

The diagram illustrates a cross-section of soil. At the top, a layer of blue water with numerous blue raindrops is shown. Below the water, the soil is depicted in a blue gradient. Three red, spherical objects labeled "Urea" are visible on the soil surface. A white arrow points from the text "Urease inhibitor" to the left of the urea spheres. In the lower right, a white arrow points from the text "Crop residue" to a dark green, irregularly shaped area representing crop residue. The soil layer below the residue is a darker blue. At the very bottom, a dark brown layer represents the subsoil.

Crop residue

Urea Urea Urea

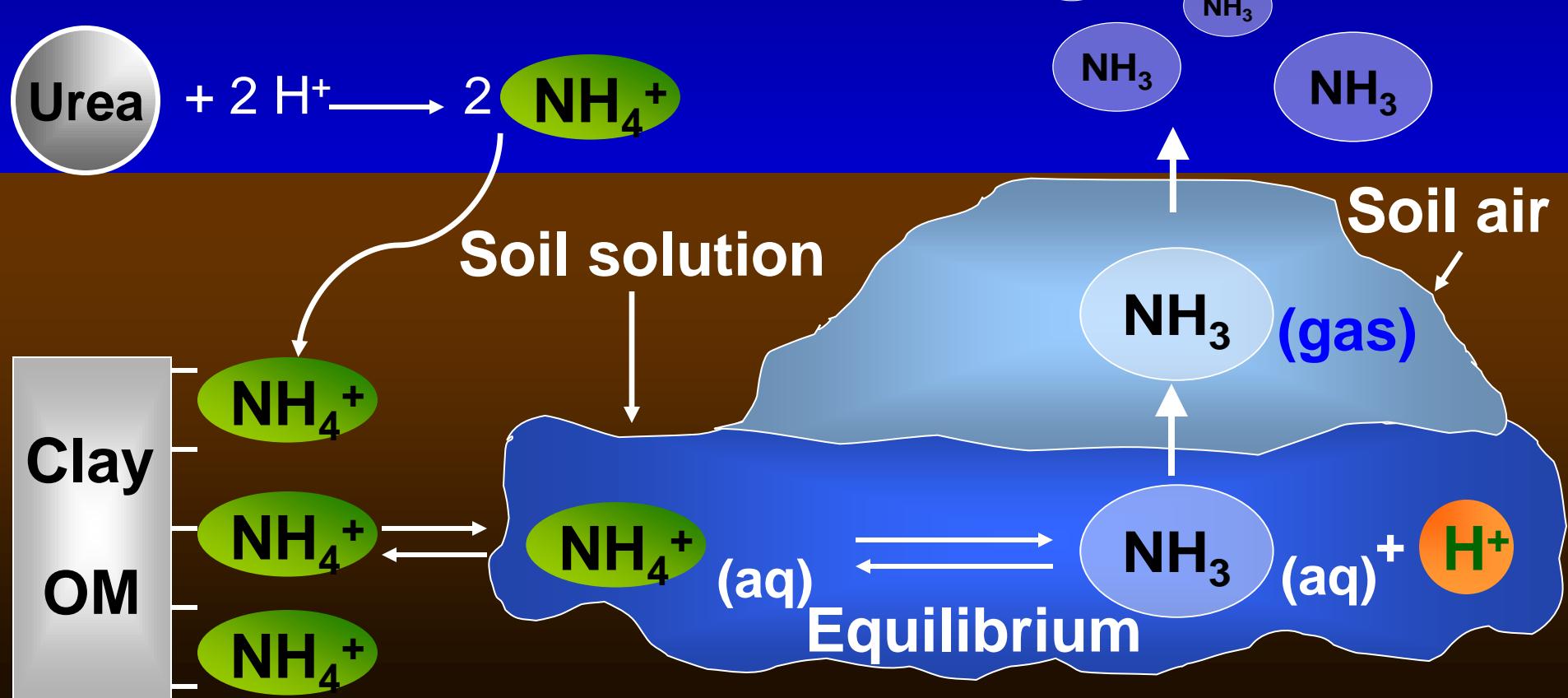
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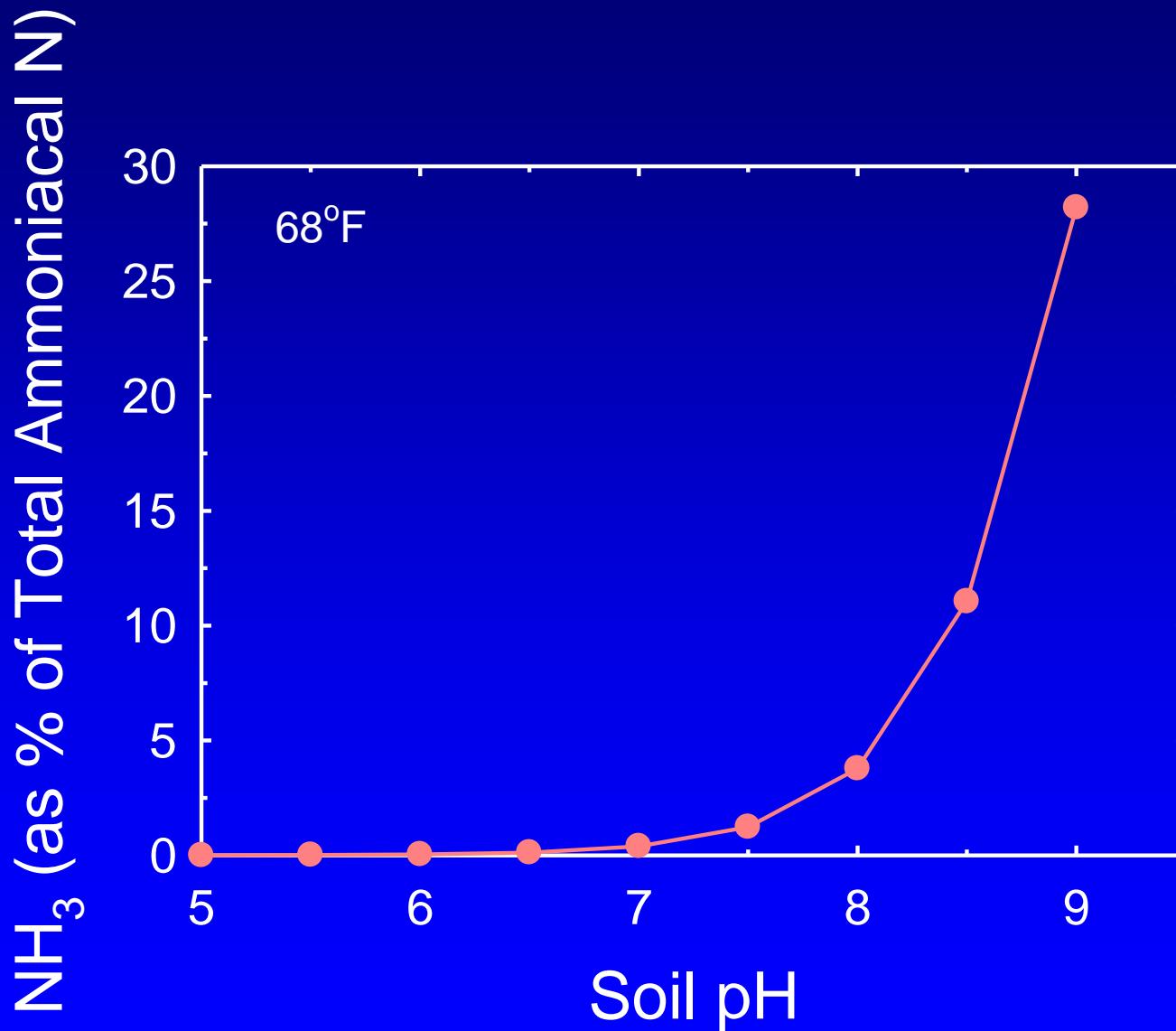
# NH<sub>3</sub> Volatilization Process

Atmosphere

Hydrolysis



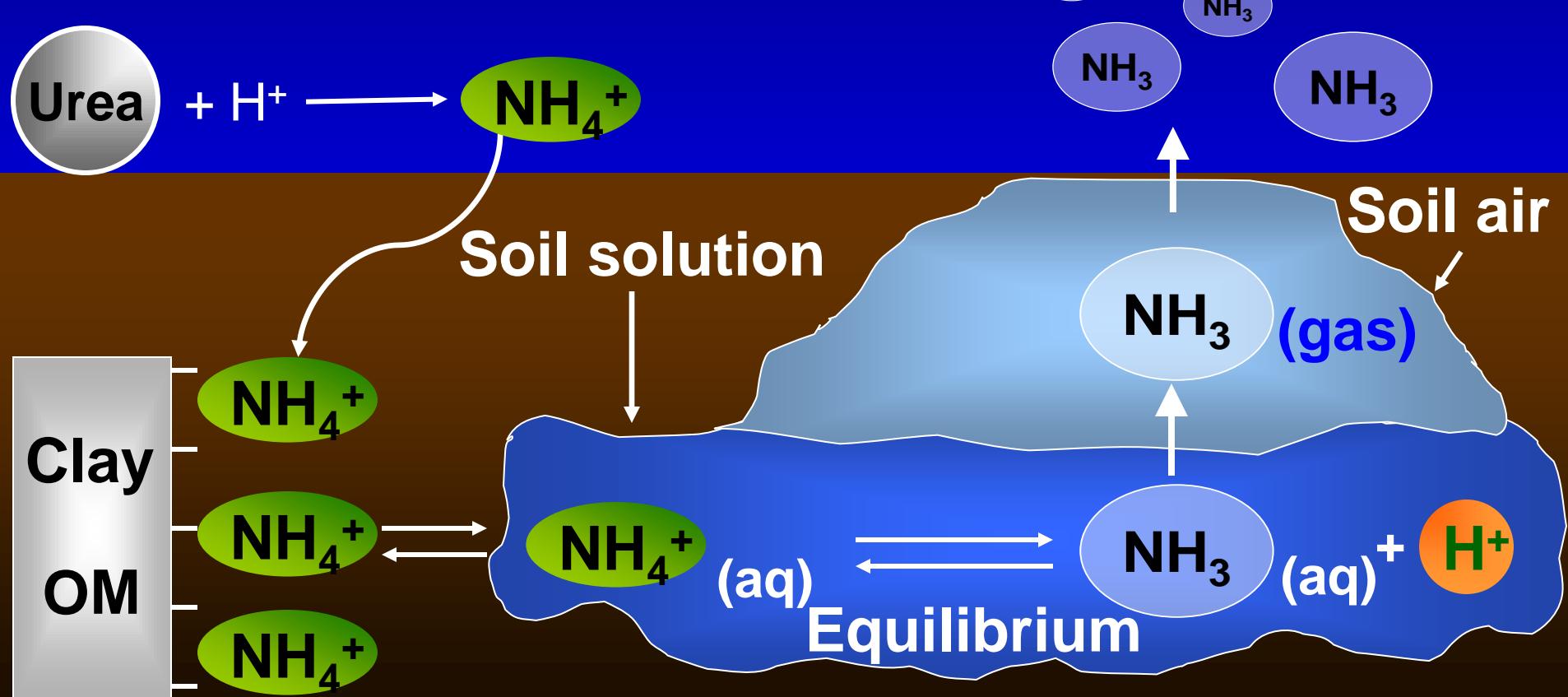
# Effect of soil pH on $\text{NH}_3$ as % of Total N



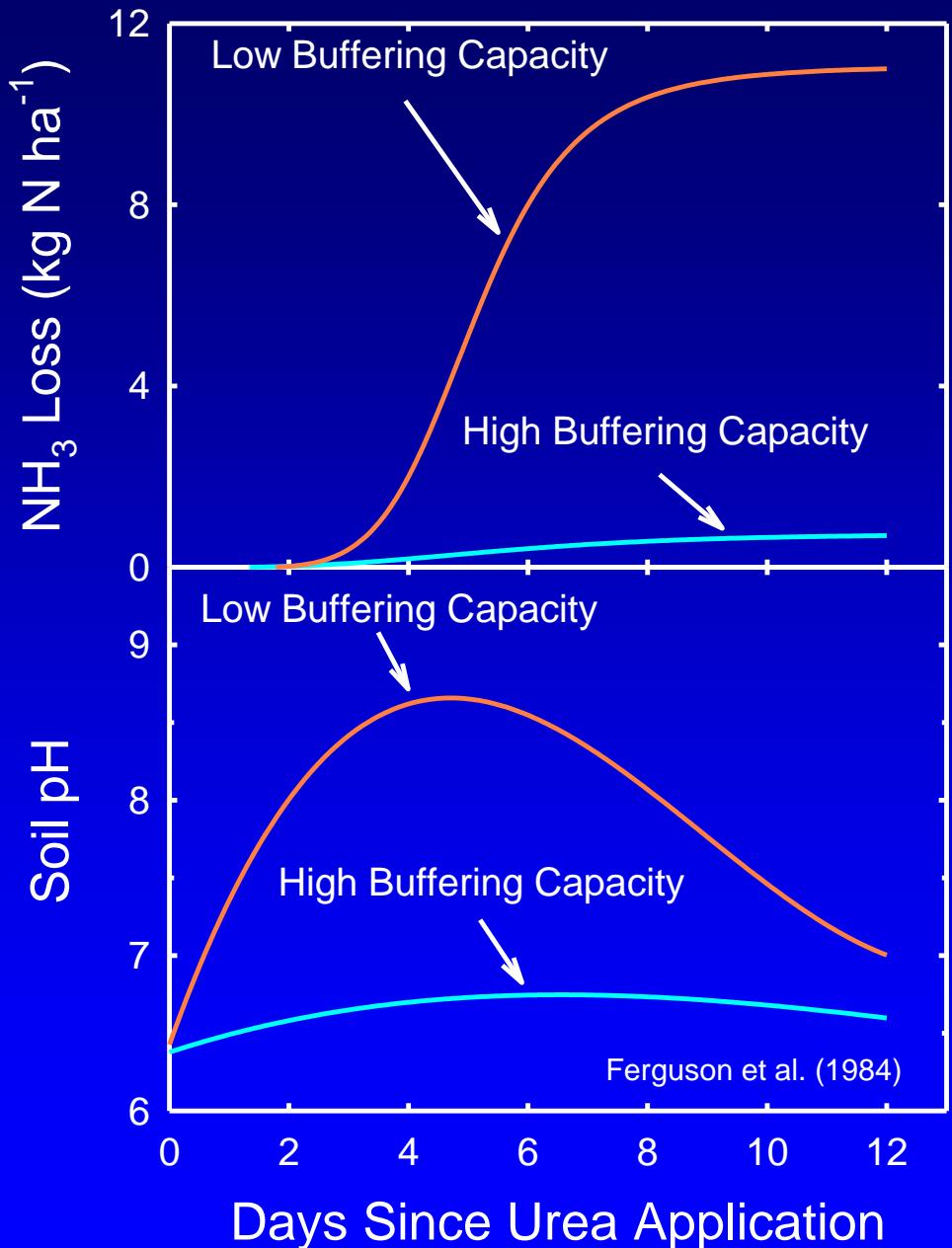
# NH<sub>3</sub> Volatilization Process

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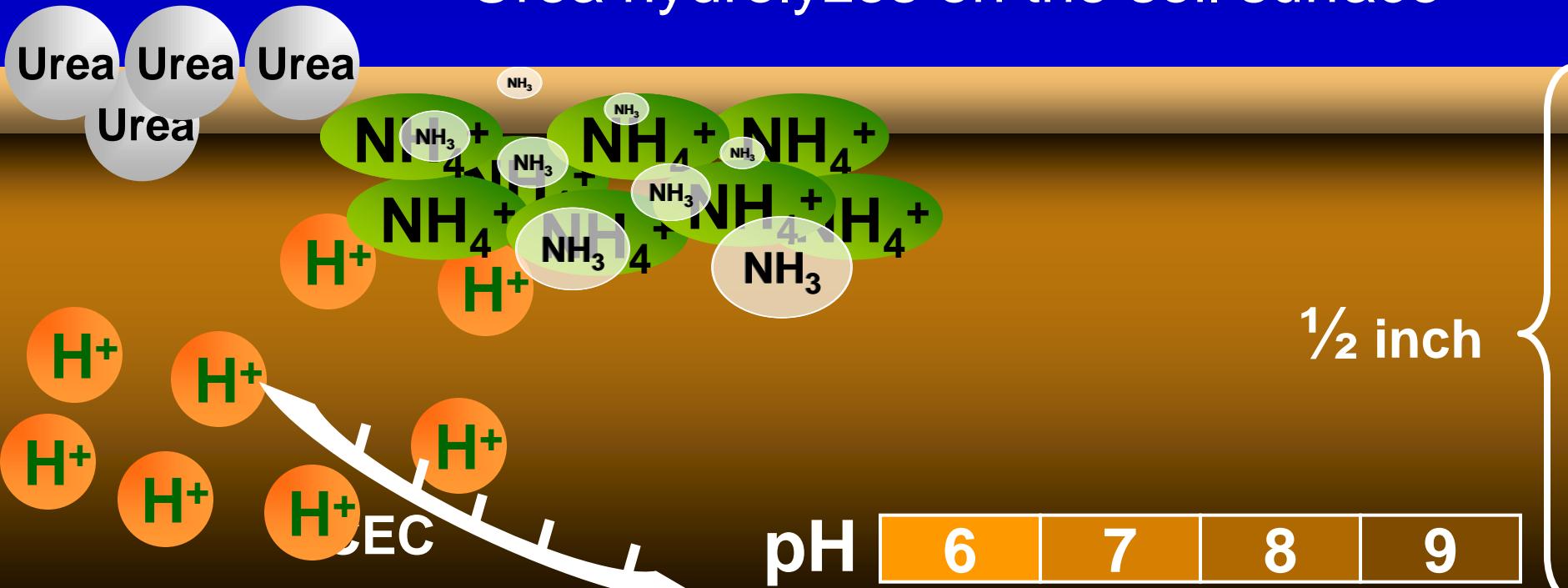


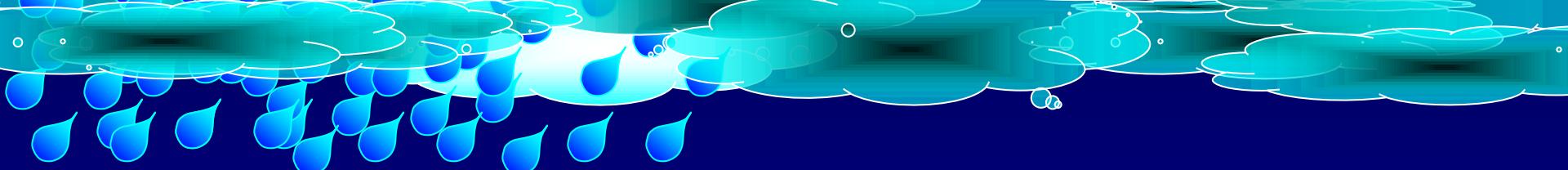
- Increase in soil pH depends on soil H<sup>+</sup> buffering capacity
- Higher clay and OM contents lead to higher soil H<sup>+</sup> buffering capacity



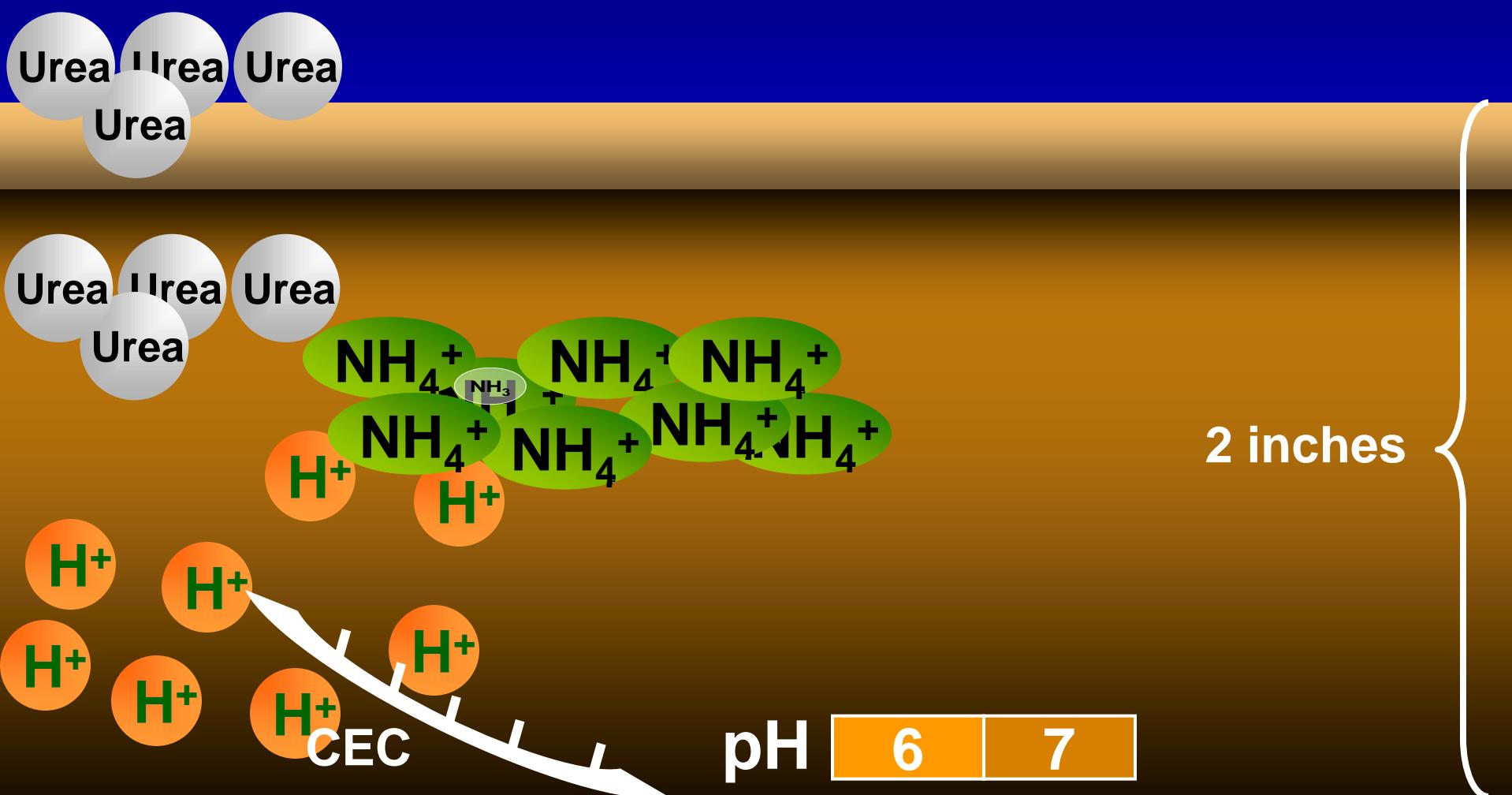
# Atmosphere

Urea hydrolyzes on the soil surface





## Urea hydrolyzes below the soil surface

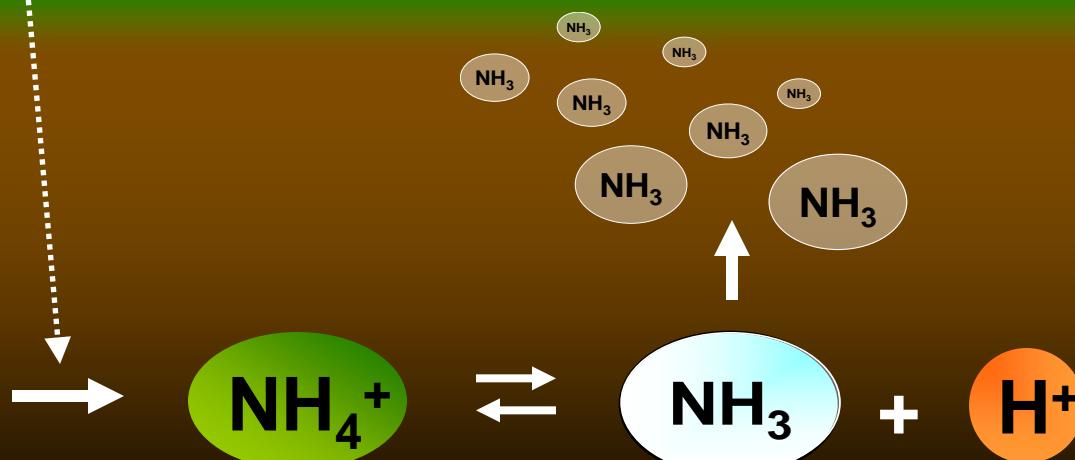


# Effect of Temperature on $\text{NH}_3$ Volatilization

Temperature

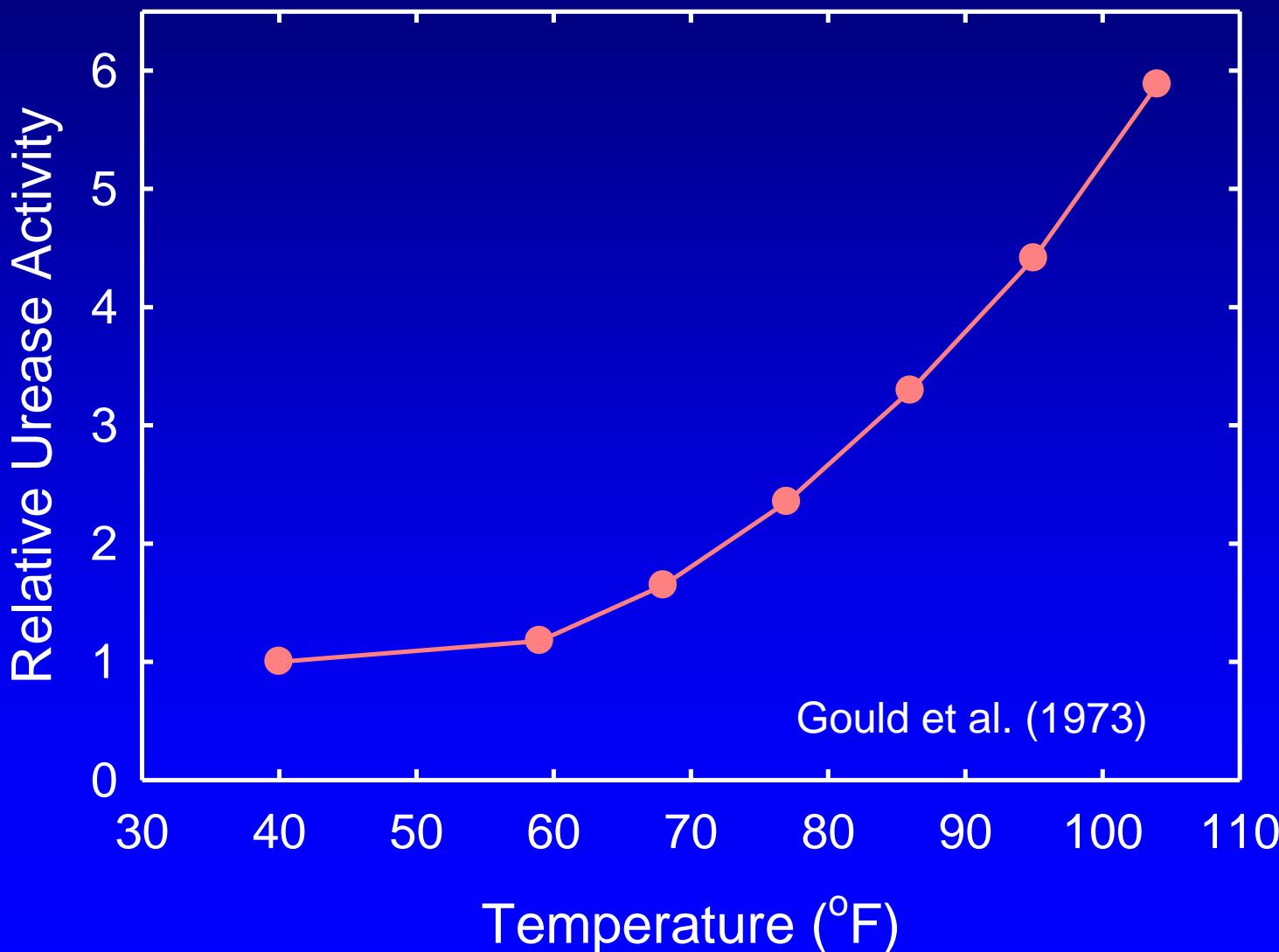
Soil

Urea



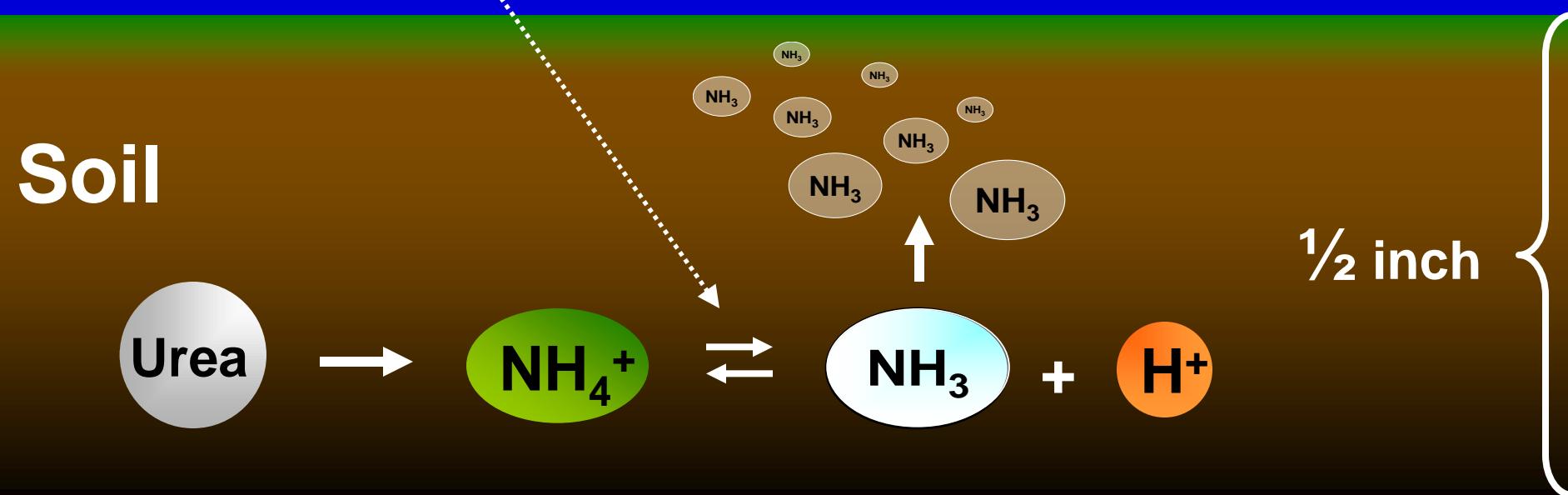
$\frac{1}{2}$  inch

# Effect of Temperature on Urea Hydrolysis

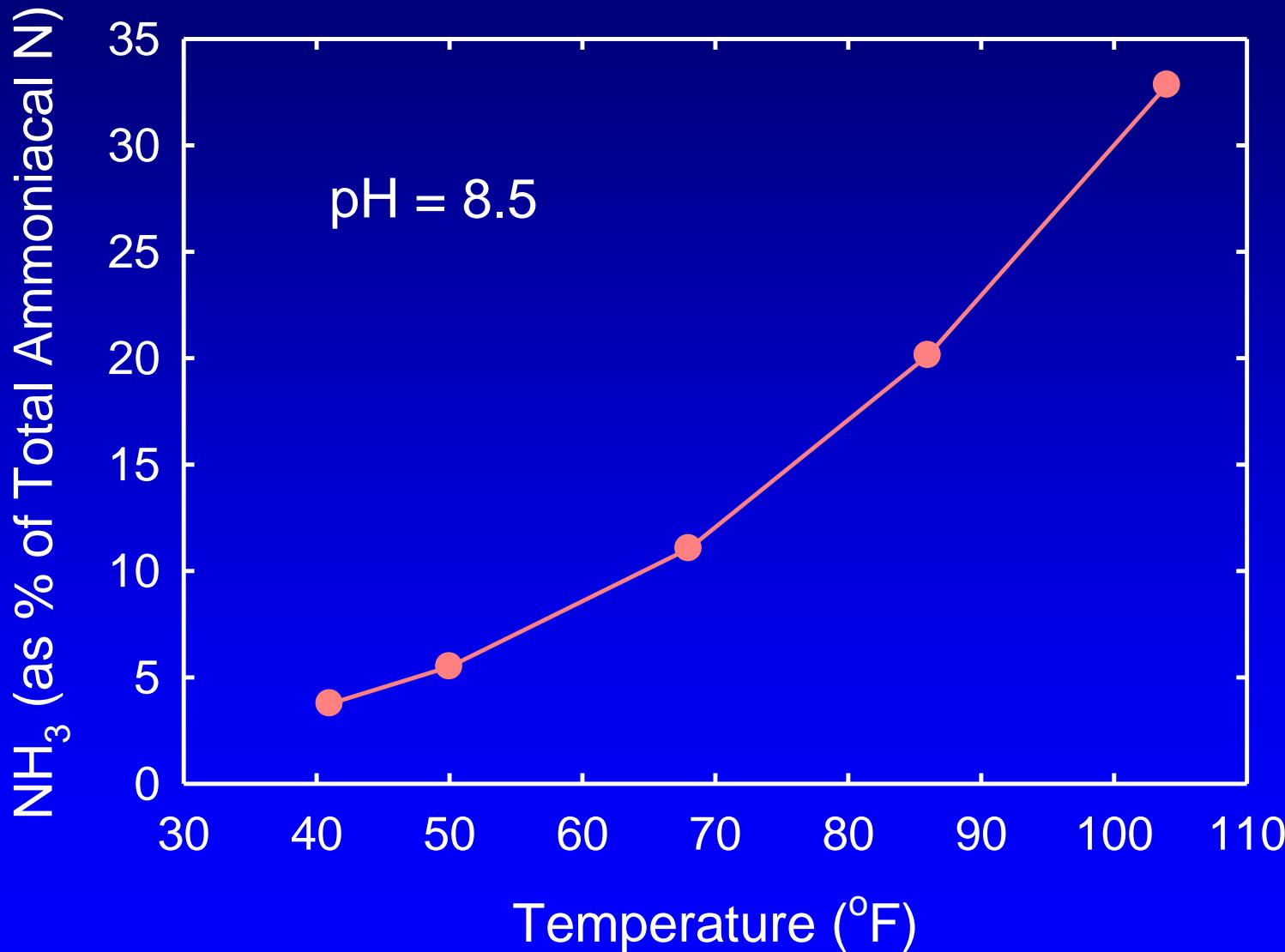


# Effect of Temperature on $\text{NH}_3$ Volatilization

Temperature

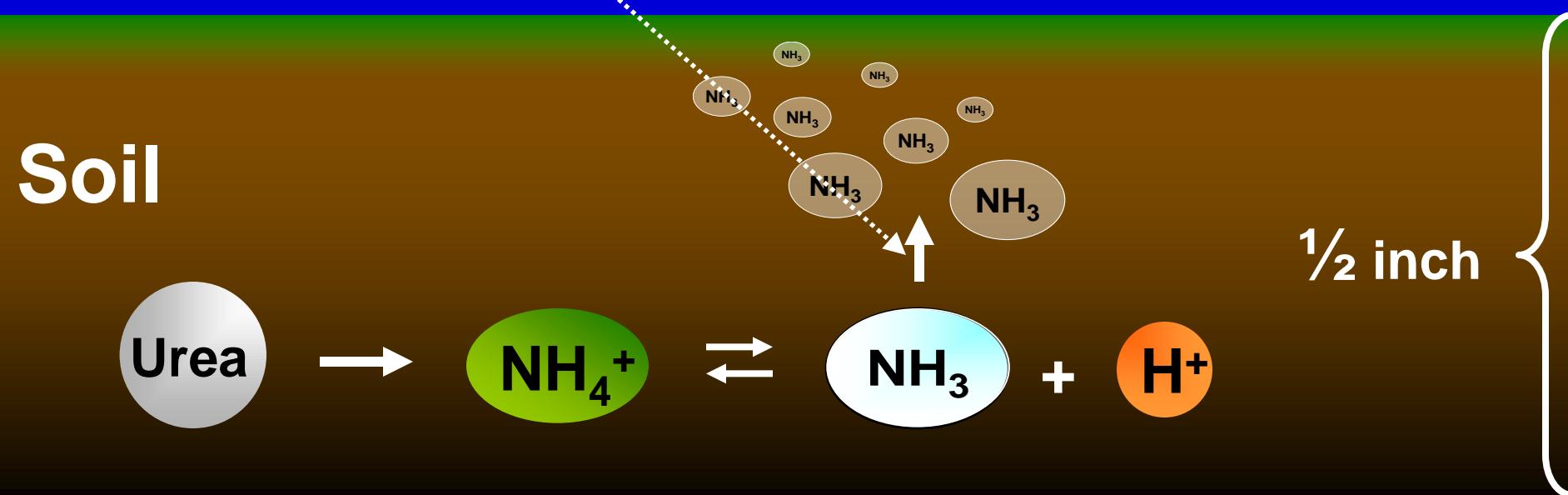


# Effect of Temperature on % of $\text{NH}_3$

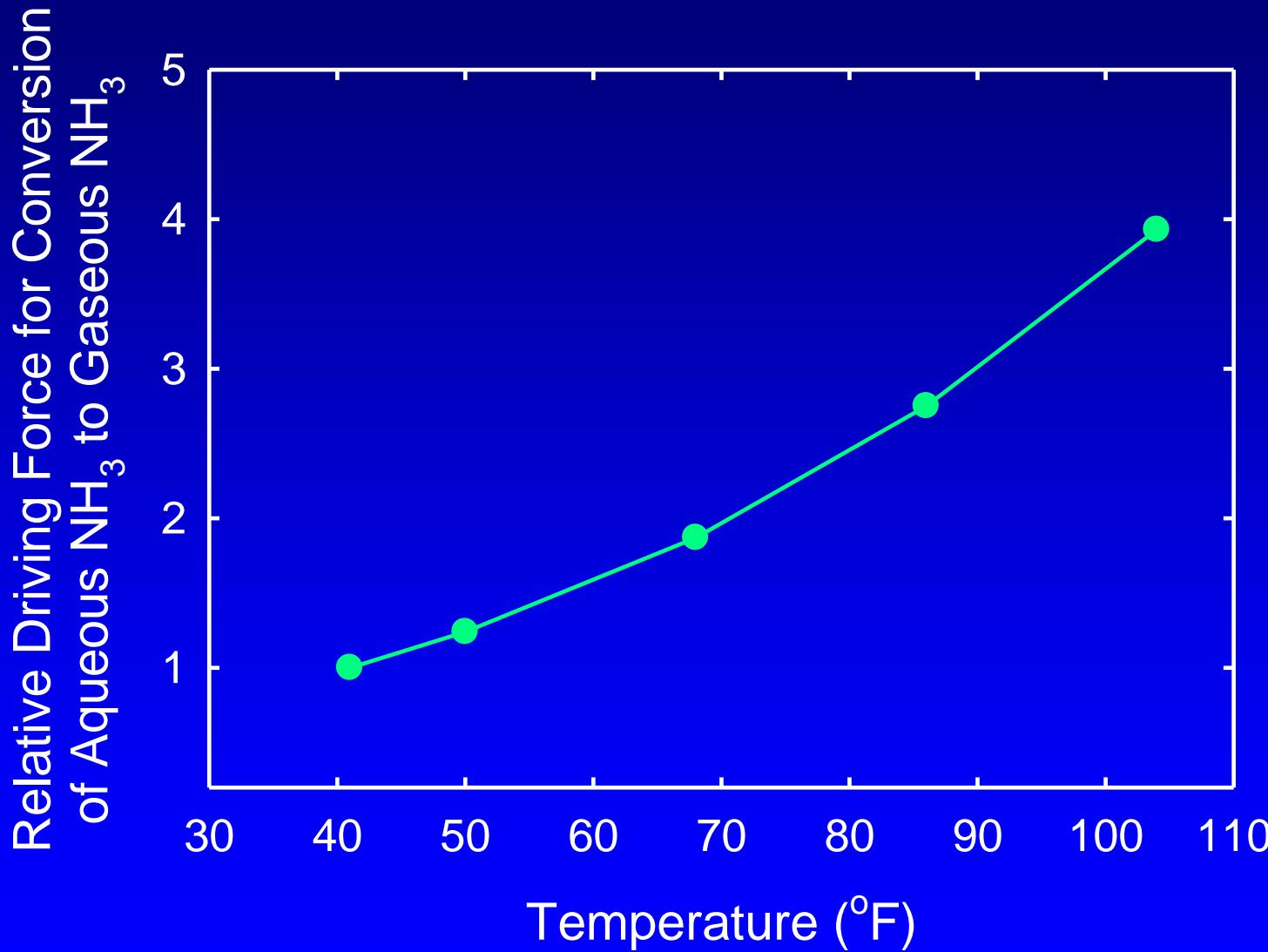


# Effect of Temperature on $\text{NH}_3$ Volatilization

Temperature

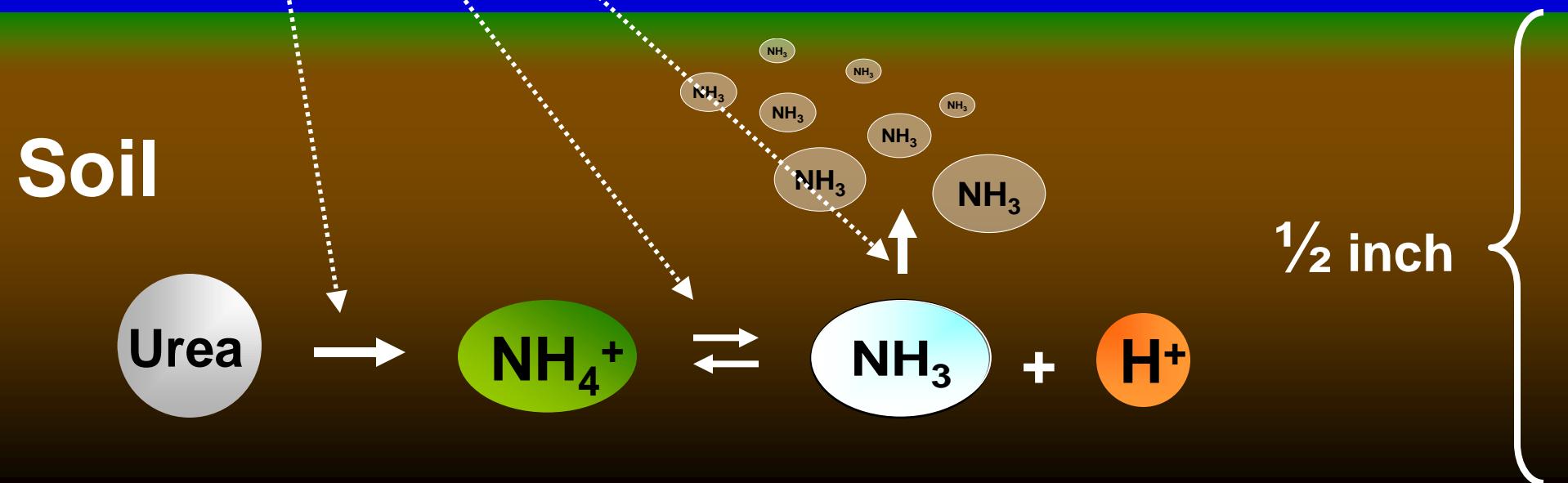


# Effect of Temperature on Conversion of Aqueous $\text{NH}_3$ to Gaseous $\text{NH}_3$



# Effect of Temperature on $\text{NH}_3$ Volatilization

Temperature



# Effect of Wind Speed on $\text{NH}_3$ Volatilization

Atmosphere

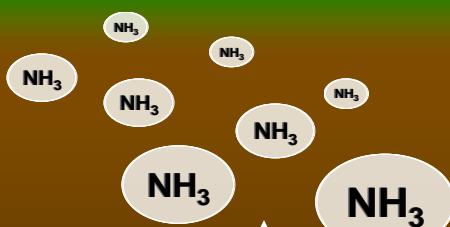
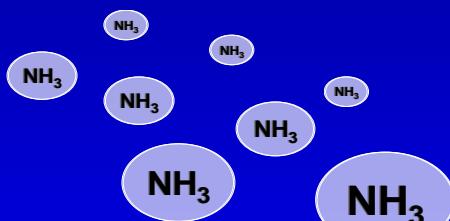
Wind speed

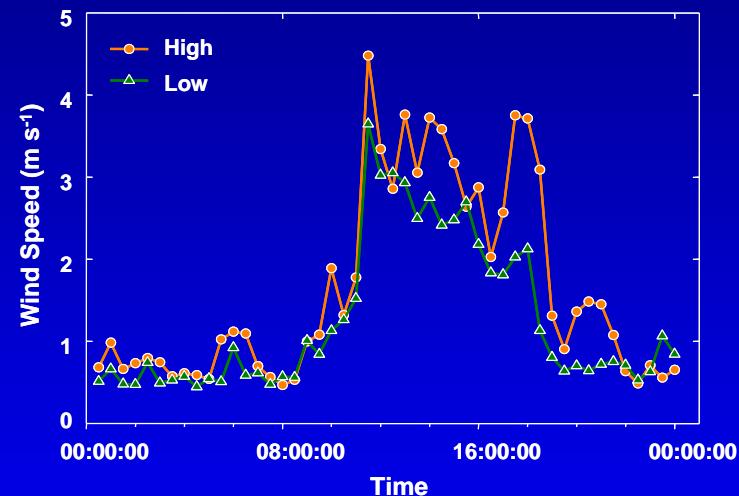
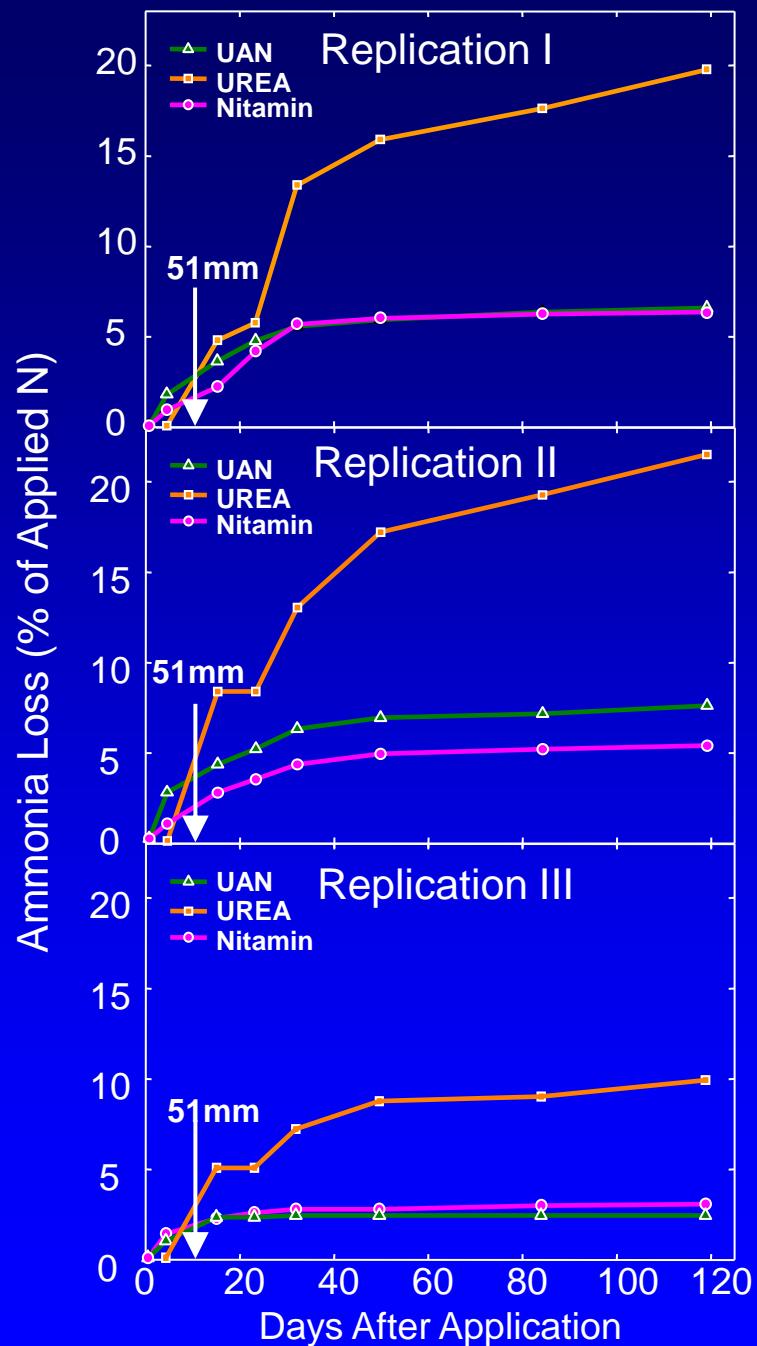


Soil



1/2 inch





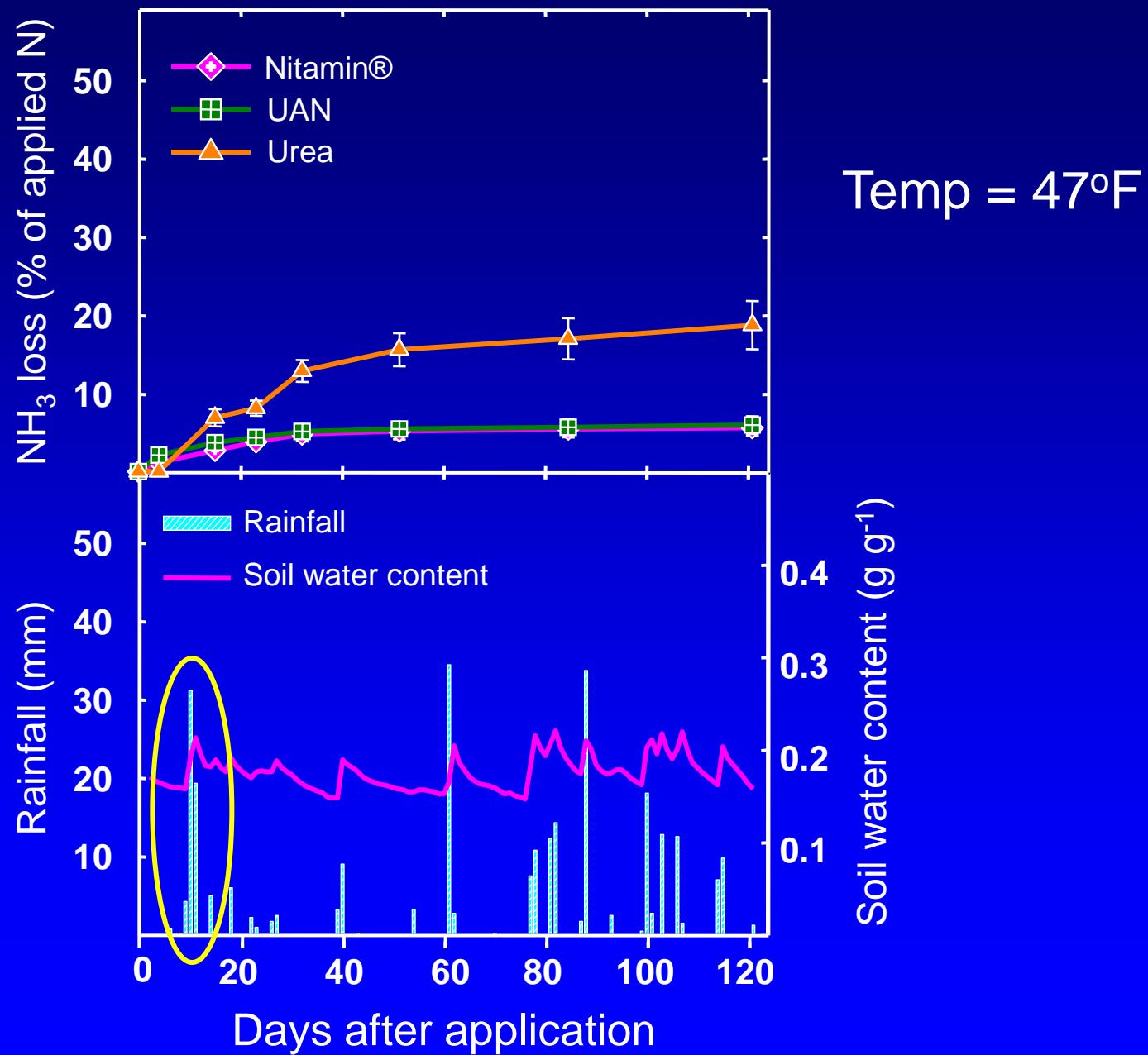
# Optimum Conditions for Volatilization

- Small diffusion depth into soil
- $\text{RH} > \text{CRH}$
- High temperature
- High wind speed
- Low pH buffering capacity

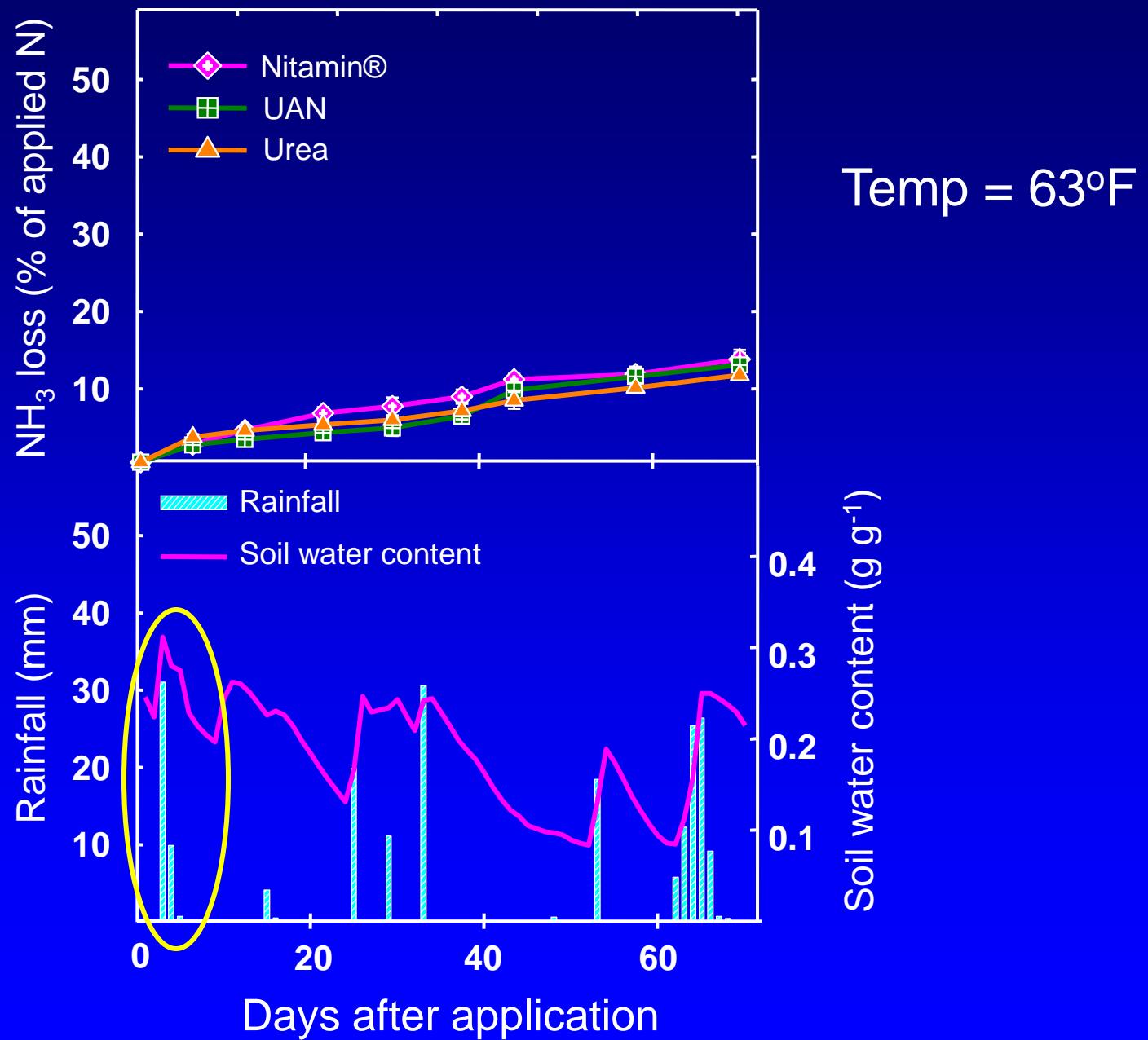
# Overview

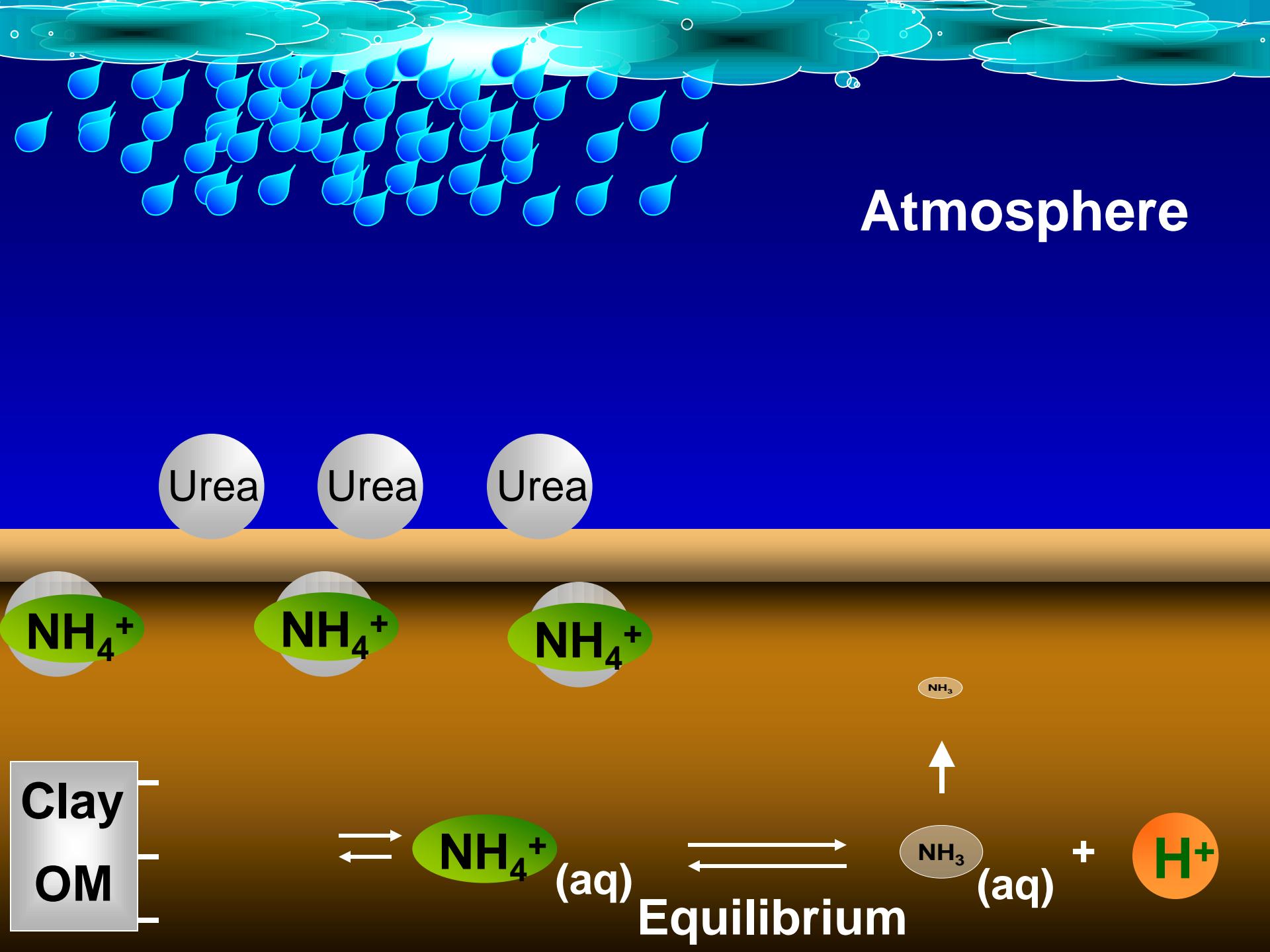
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# Urea Applied to Pasture - Fall 2004



# Urea Applied to Pasture – Spring 2005





# Atmosphere

Urea

Urea

Urea

$\text{NH}_4^+$

$\text{NH}_4^+$

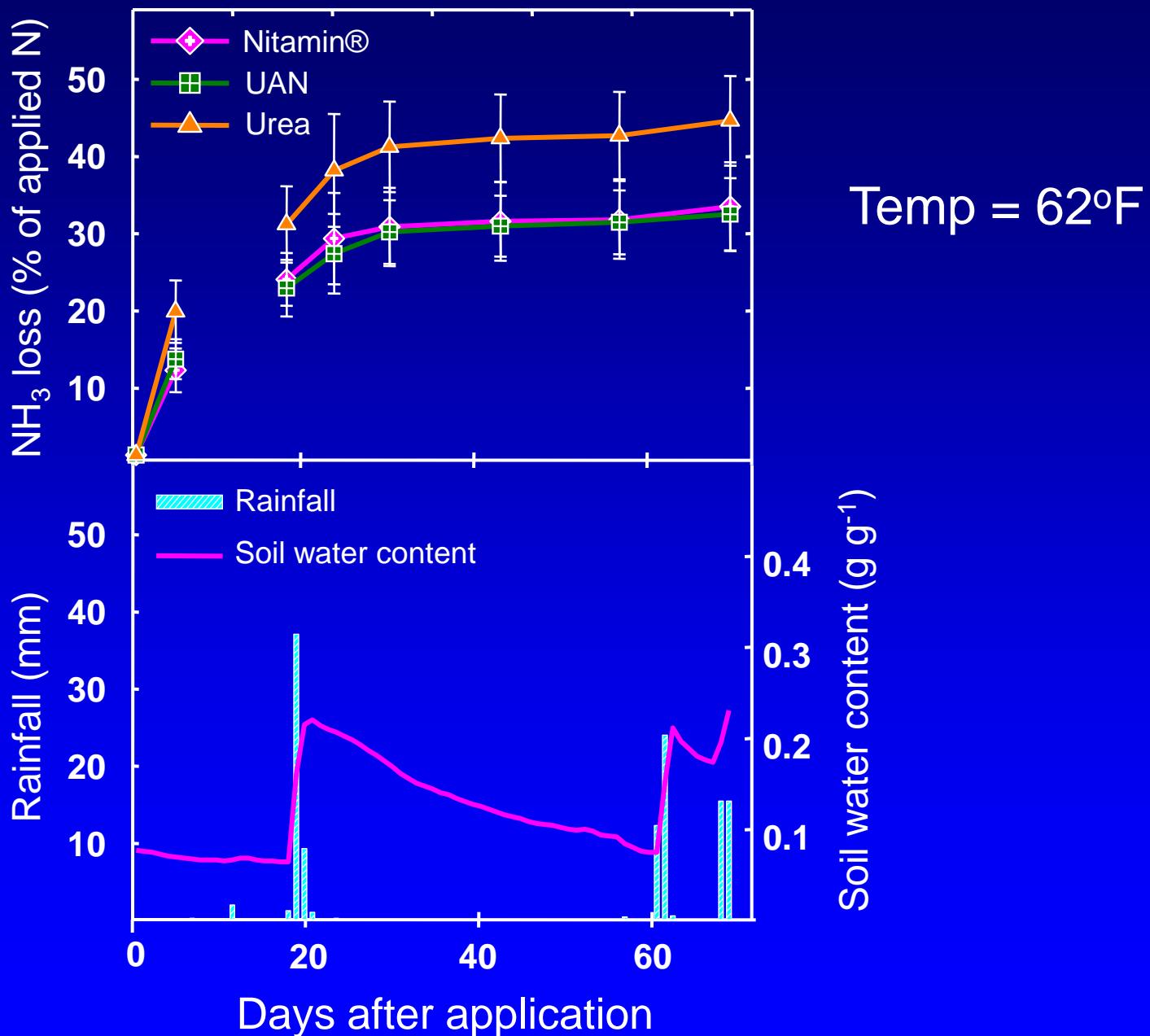
$\text{NH}_4^+$

$\text{NH}_3$

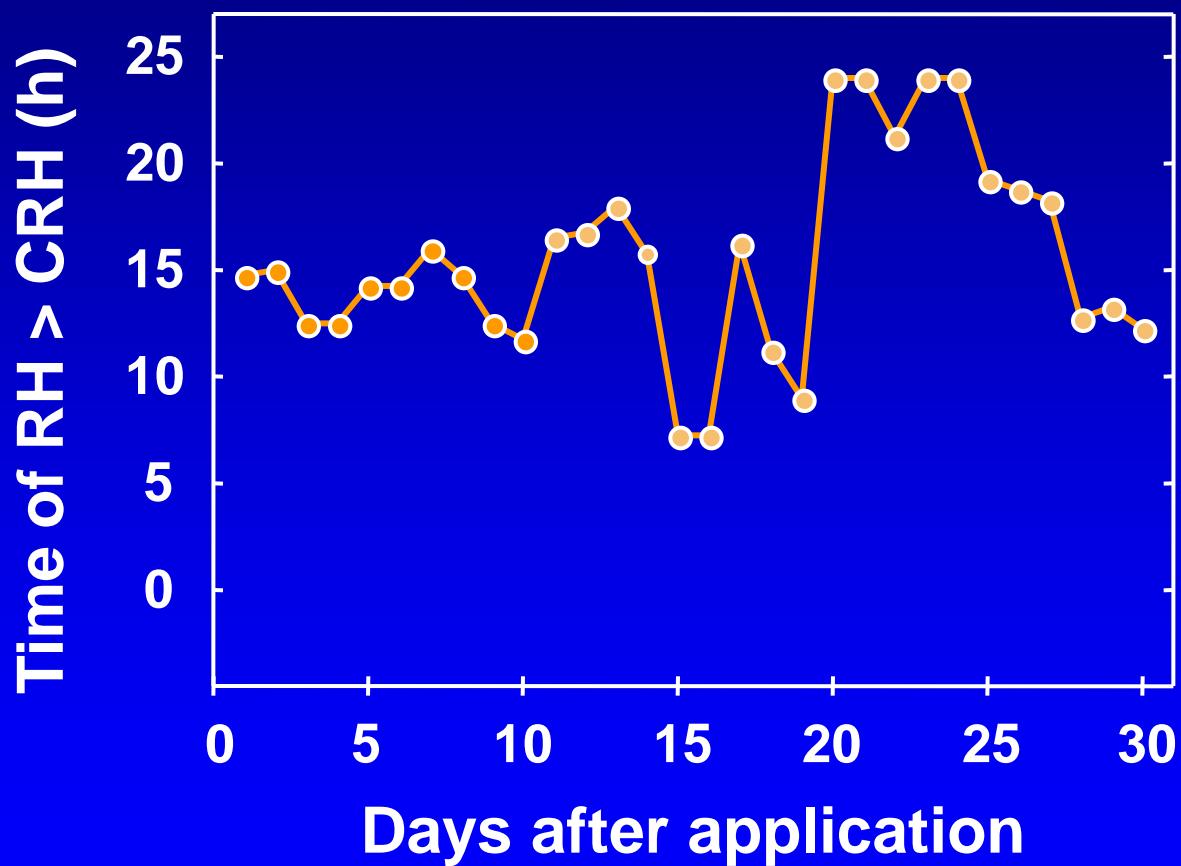
Clay  
OM



# Urea Applied to Pasture - Fall 2005



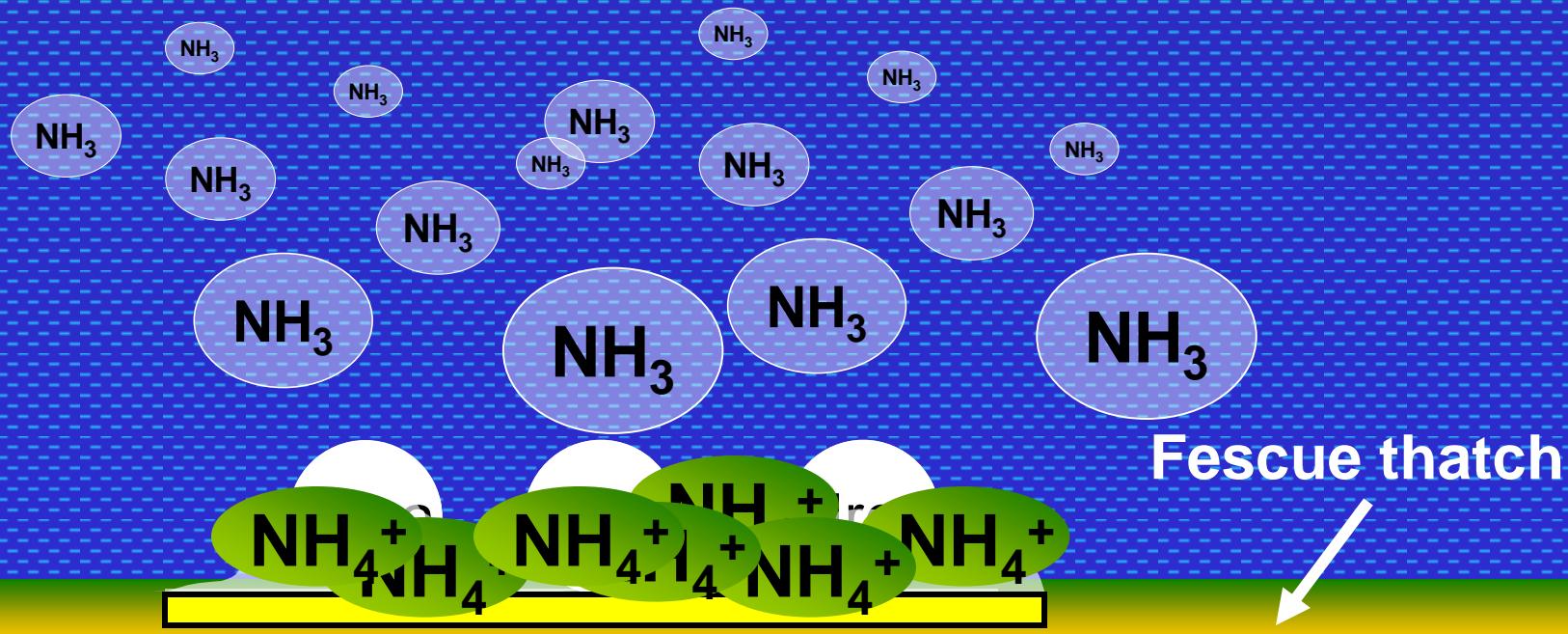
# Time of RH above CRH in the first 30 d



Total: RH > CRH = 183 days

# Atmosphere

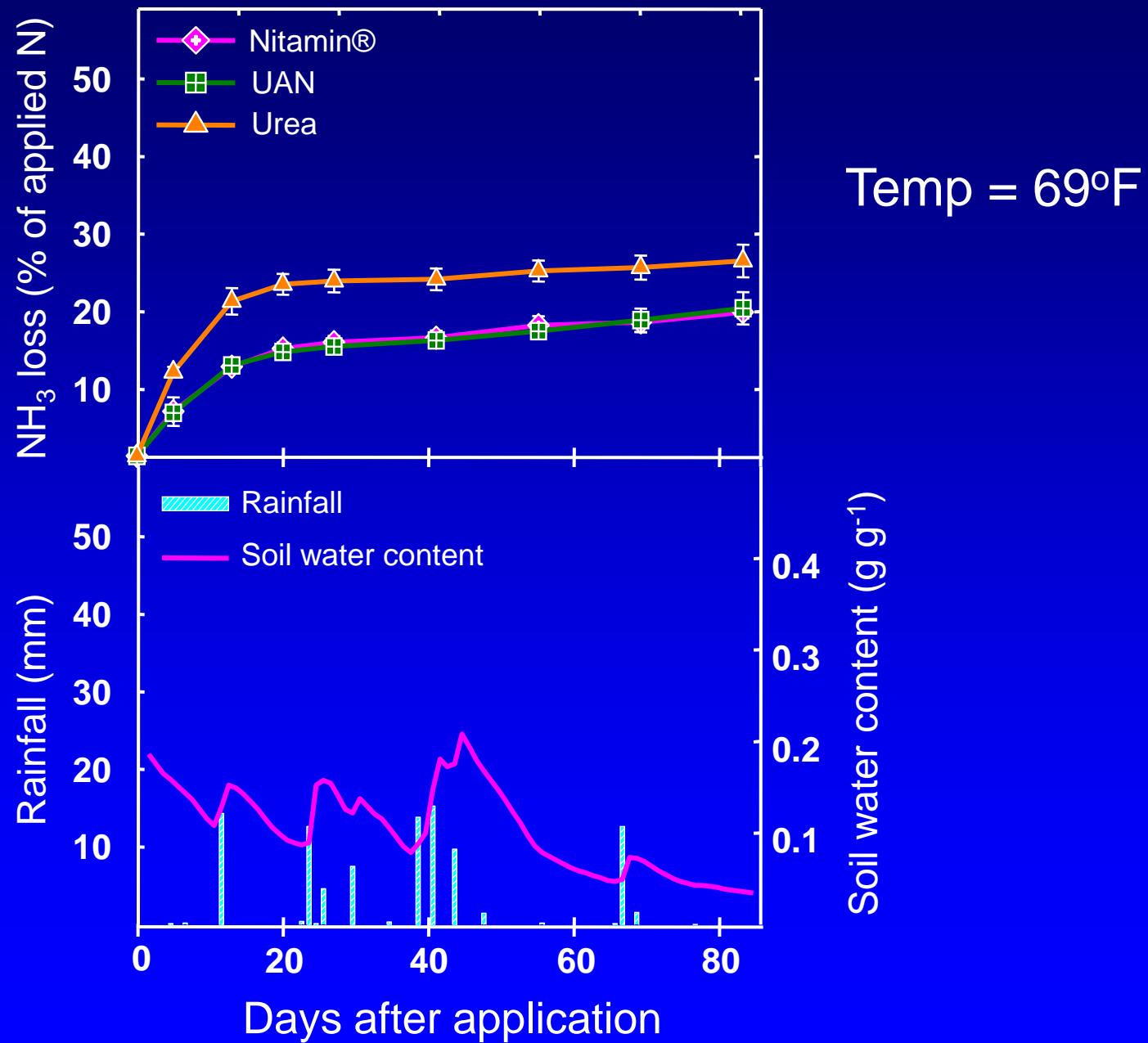
$\text{RH} > \text{CRH} (80\%)$



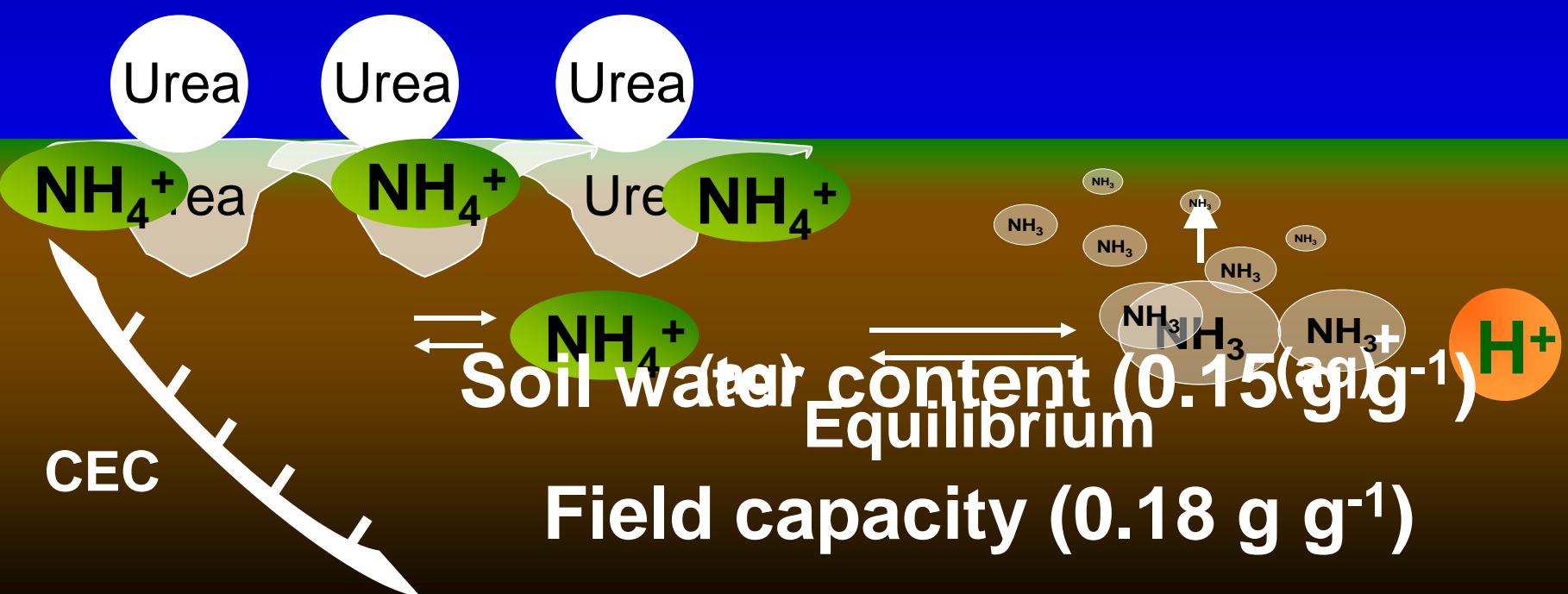
Soil water content ( $0.07 \text{ g g}^{-1}$ )

Permanent wilting point ( $0.04 \text{ g g}^{-1}$ )

# Urea Applied to Pasture – Spring 2006



# Atmosphere



# Summary of Pasture Results

## Ammonia volatilization losses under field conditions

Fertilizer	Fall 2004	Spring 2005	Fall 2005	Spring 2006
----- Ammonia loss (% of applied N) -----				
Urea	19 a*	12 a	46 a	24 a
UAN	6 b	13 a	33 b	18 a
Nitamin®	6 b	14 a	34 b	18 a

•Within a column, values followed by the same letter are not significantly different according to Fisher's LSD at p=0.05

# Summary for Pastures/No-Till

Dry Soil, Dry Residue, RH < CRH: OK

Dry Soil, Dry Residue, RH > CRH: not OK

Wet Soil, Wet Residue: not OK

- Apply on dry soil, dry residue, expected low RH, and hope for 1 inch of rain whenever it rains.
- Apply on wet soil if 1 inch rain/irrigation is expected.

# Summary for Clean-Till Crops

Dry Soil: OK

Wet Soil: not OK

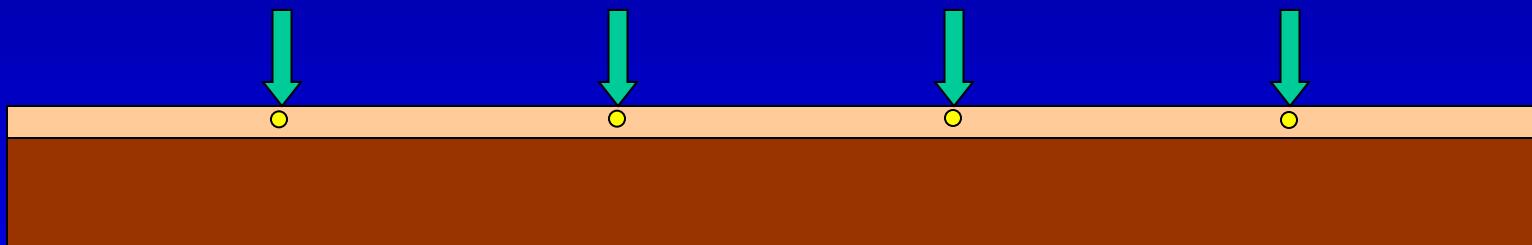
- Apply on dry soil and hope for 1 inch of rain whenever it rains.
- Apply on wet soil if 1 inch of rain/irrigation is expected.

# Summary for Pastures/Crops

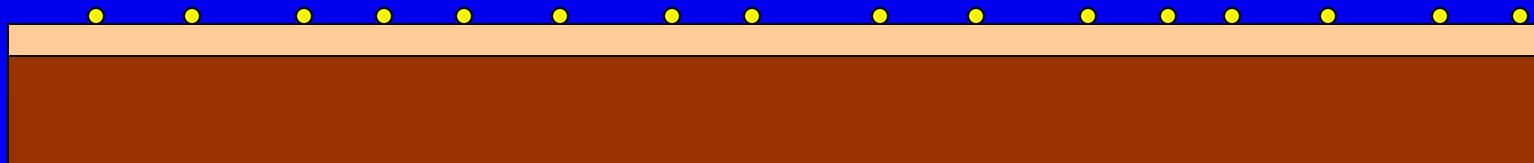
Knifed or disked in



Dribbled

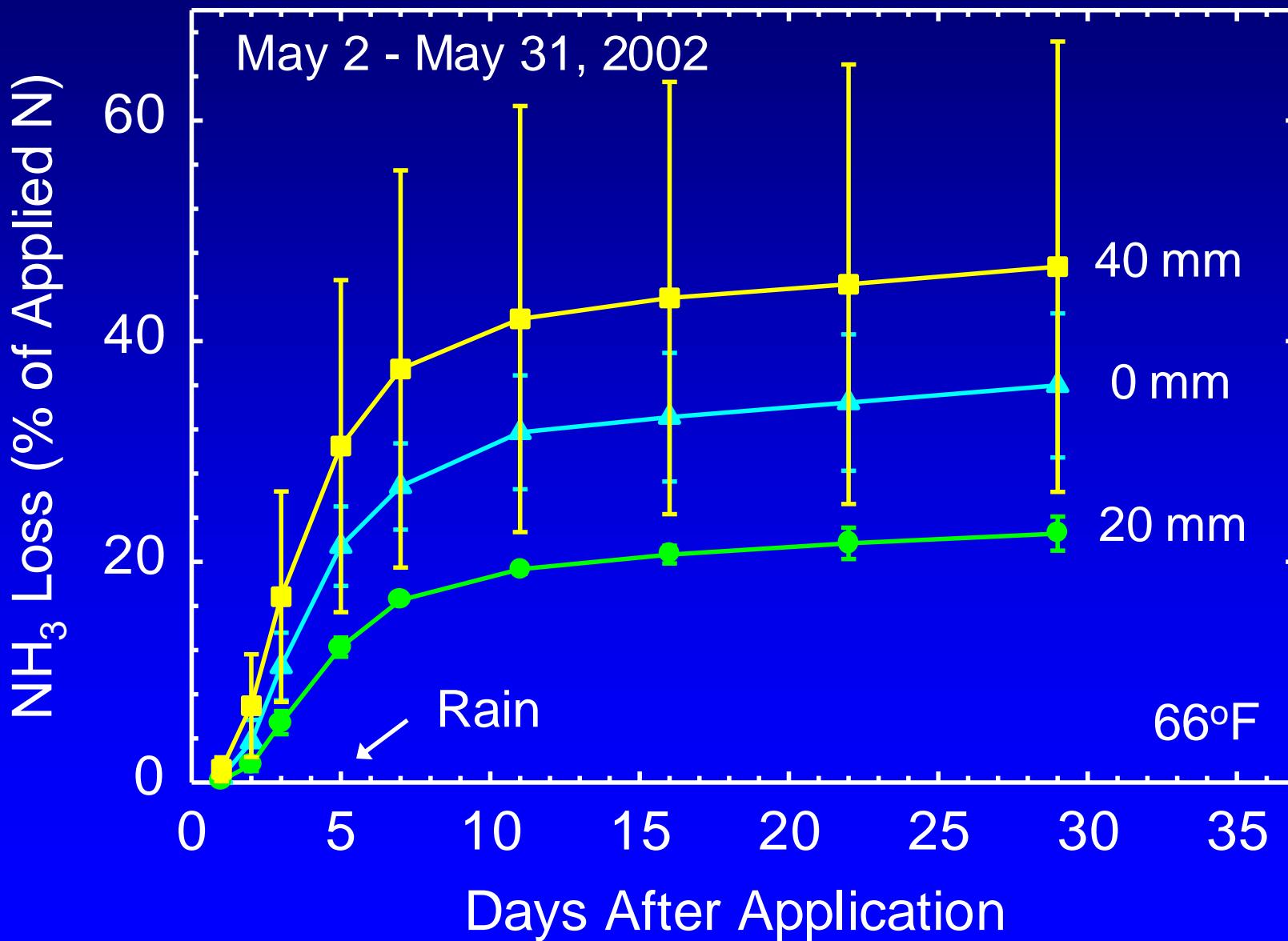


Broadcast

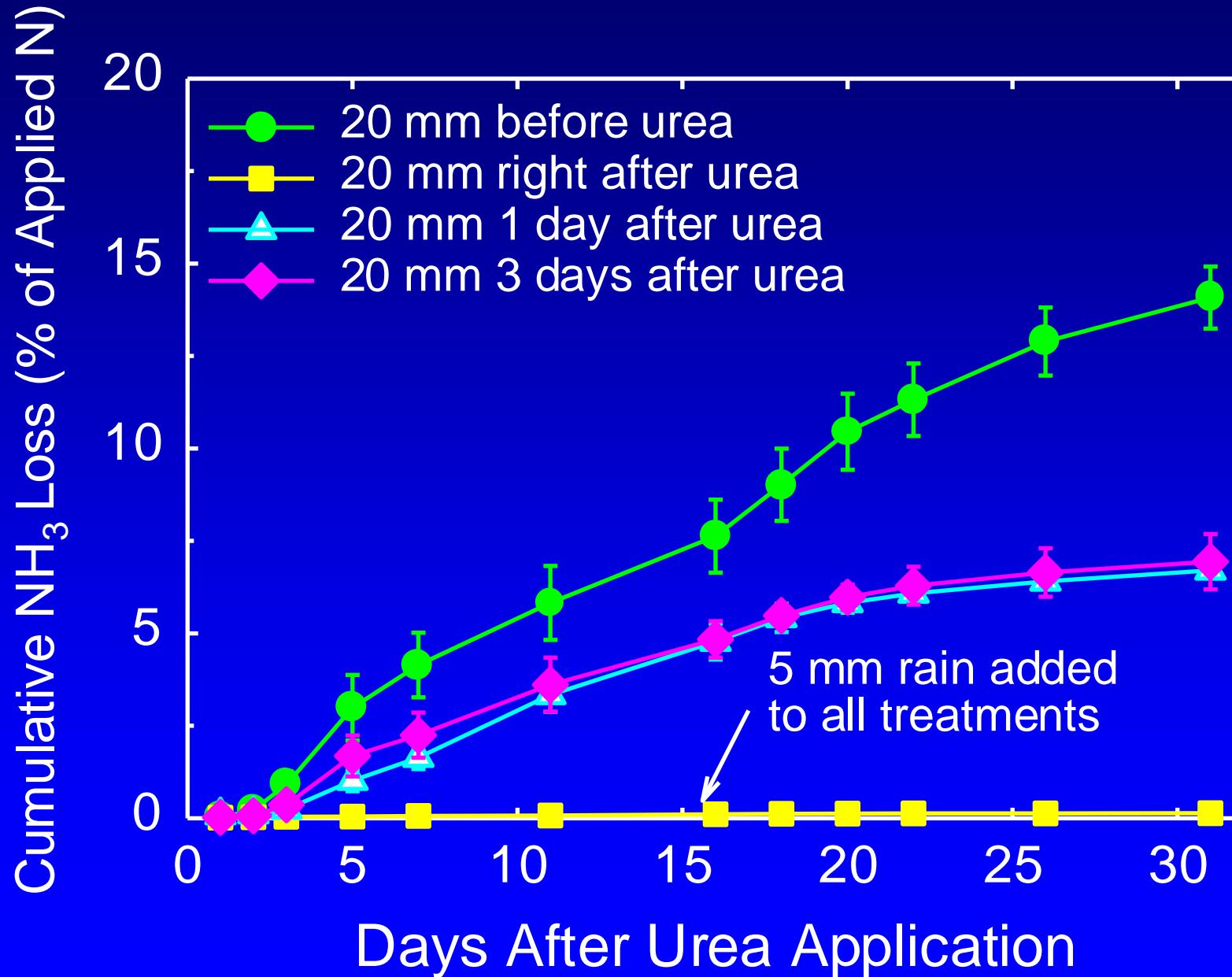


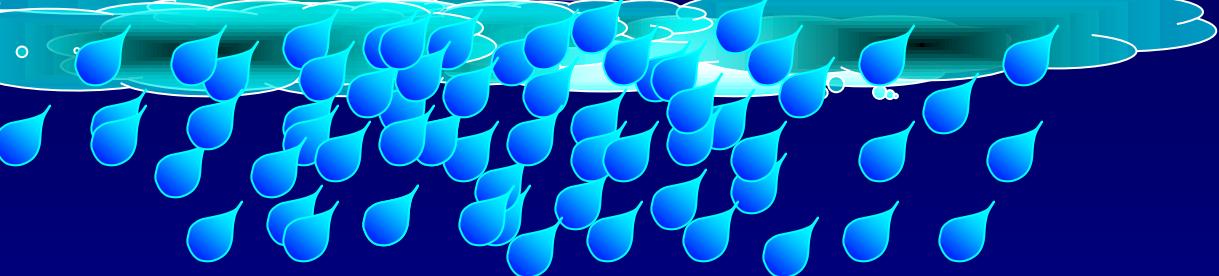
- Knifed or disked in better than Dribbled
- Dribbled better than Broadcast

# Urea Applied to Pine Plantations

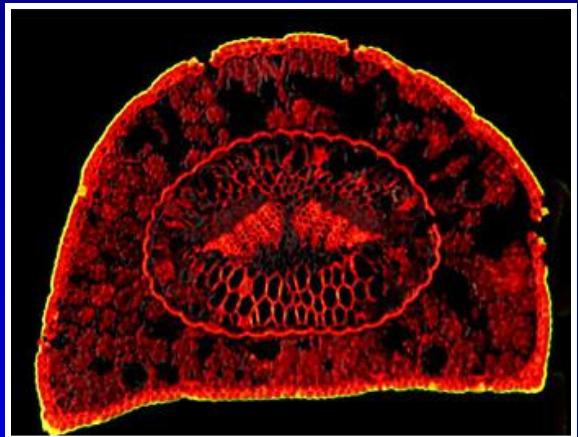


# Laboratory Results (77°F, 90% RH)

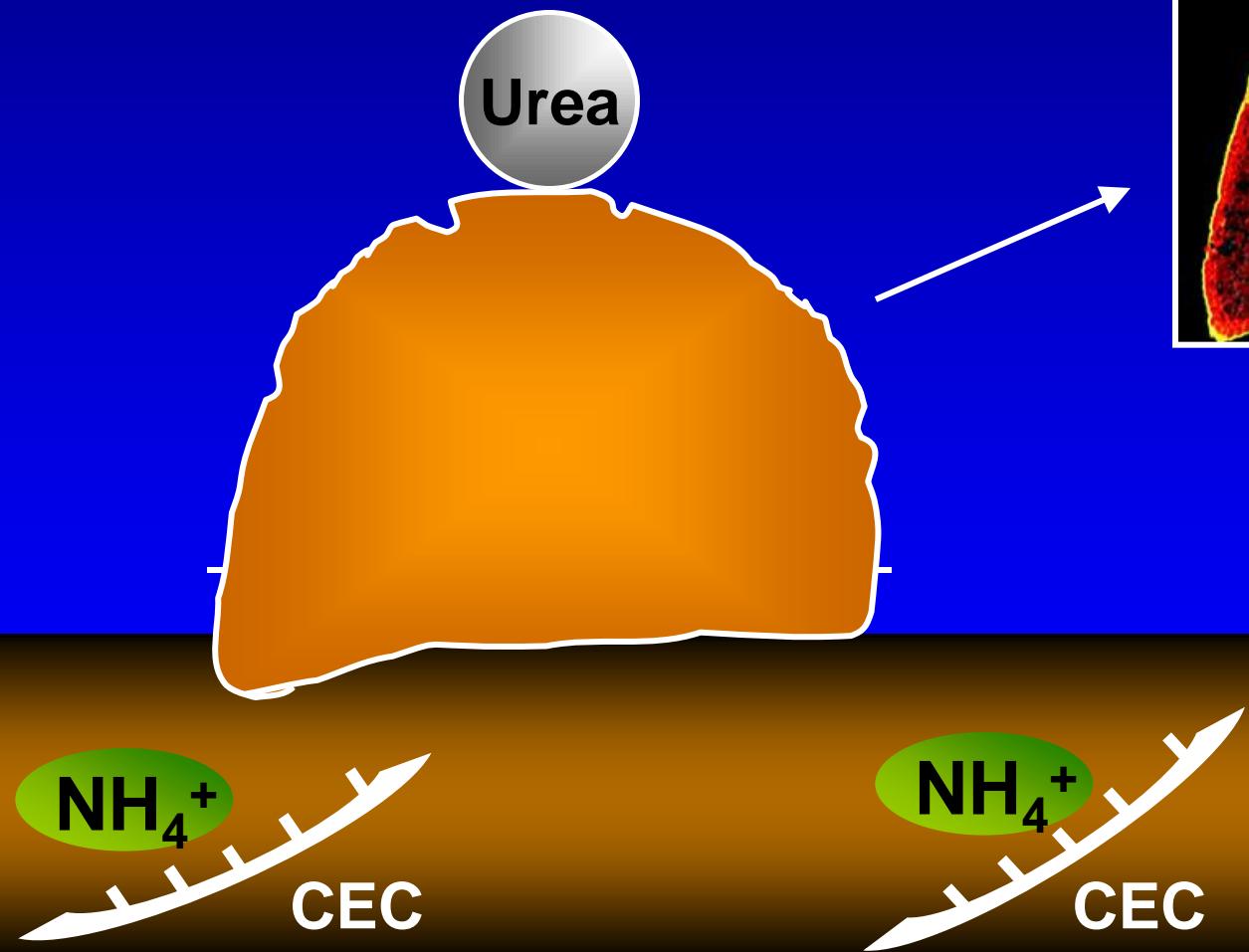


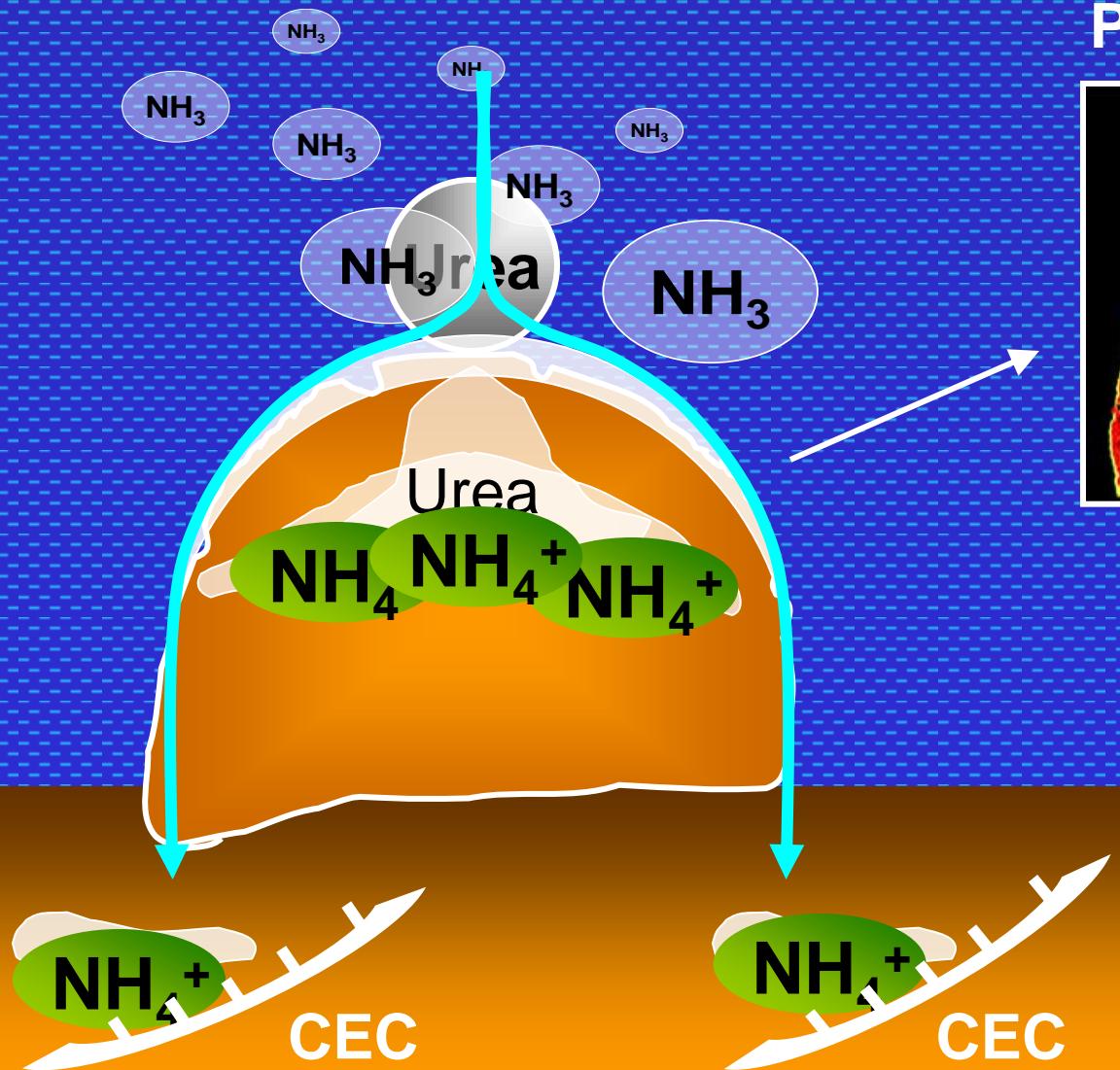
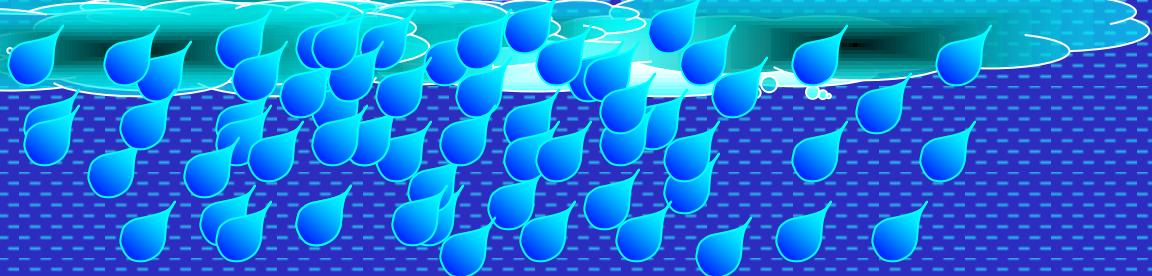


Pine needle photo

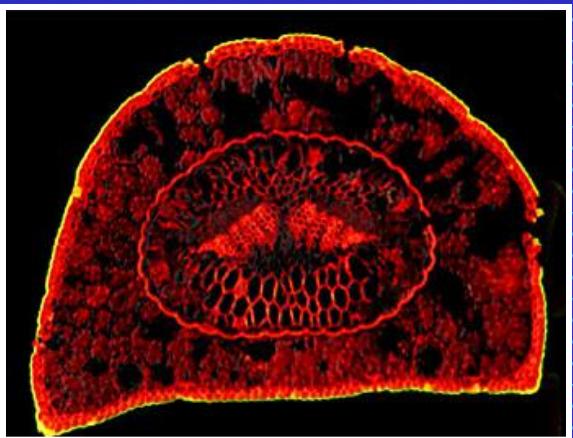


[www.microscopyu.com](http://www.microscopyu.com)





Pine needle photo



[www.microscopyu.com](http://www.microscopyu.com)

Organic Layer

# Summary for Pine Plantations

Dry Forest Floor, RH < CRH: OK

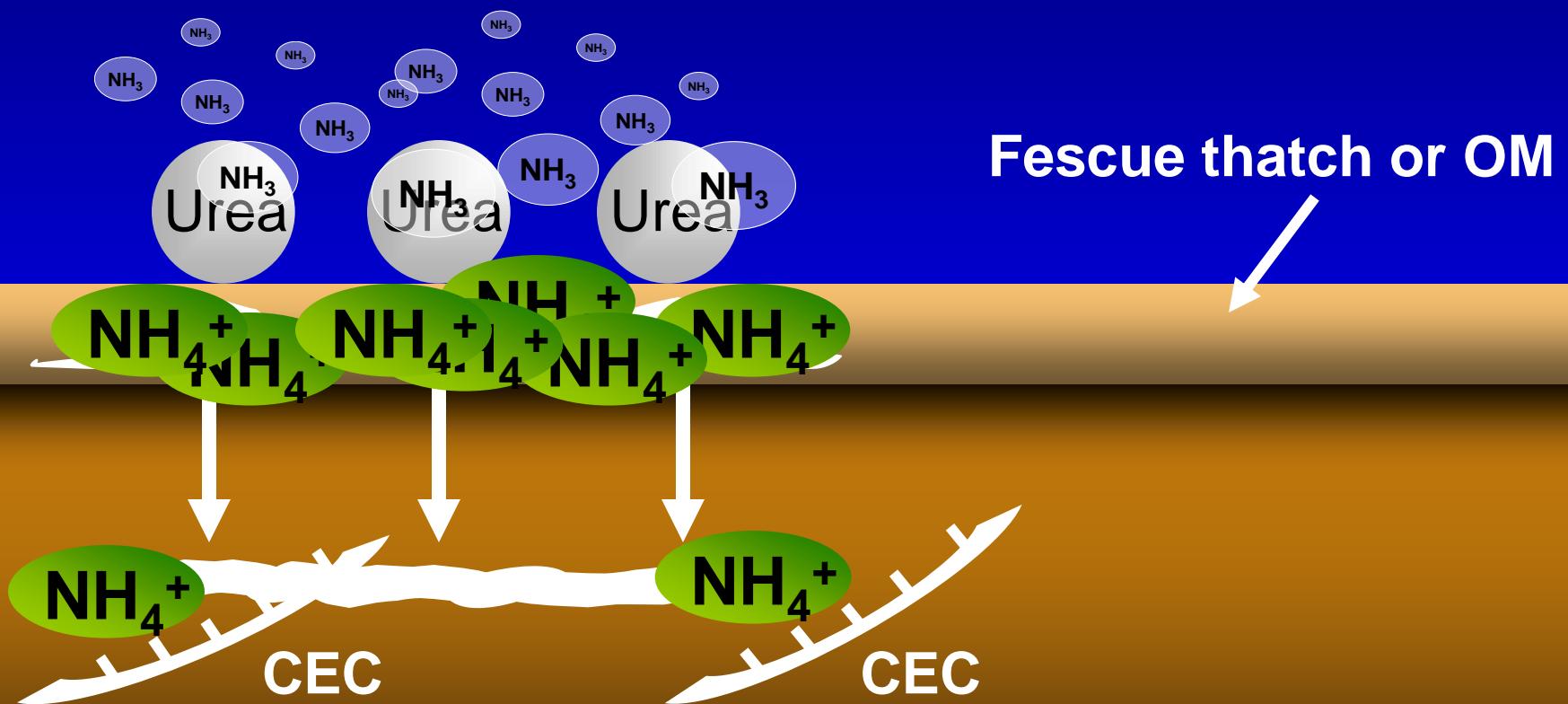
Dry Forest Floor, RH > CRH: not OK

Wet Forest Floor: not OK

- Apply on dry forest floor, with expected low RH, and hope for 1 inch of rain whenever it rains.
- Apply on wet forest floor if 1 inch rain is expected.

# Questions?

# Atmosphere



# Laboratory Results (77°F, 90% RH)

## NH<sub>3</sub> volatilization in the lab

